

A multiscale modelling study on the sense and nonsense of thermal conductivity enhancement of liquid chromatography packings and other potential solutions for viscous heating effects

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Published in:
Journal of Chromatography A

DOI:
[10.1016/j.chroma.2020.461022](https://doi.org/10.1016/j.chroma.2020.461022)

Publication date:
2020

License:
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Document Version:
Accepted author manuscript

[Link to publication](#)

Citation for published version (APA):
Deridder, S., Smits, W., Broeckhoven, K., & Desmet, G. (2020). A multiscale modelling study on the sense and nonsense of thermal conductivity enhancement of liquid chromatography packings and other potential solutions for viscous heating effects. *Journal of Chromatography A*, 1620, [461022].
<https://doi.org/10.1016/j.chroma.2020.461022>

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Highlights

Multilevel simulation study of viscous heating effects at pressures up to 2500bar

Effective conductivity values calculated for the materials most commonly used in LC

Halving radial temperature difference requires doubling the bed conductivity

Only structures with connected enhanced conductivity cores can achieve this

Ultra-low conductivity column walls can alleviate viscous heating effects as well

1 **A Multiscale Modelling Study on the Sense and Nonsense of Thermal**
2 **Conductivity Enhancement of Liquid Chromatography Packings and Other**
3 **Potential Solutions for Viscous Heating Effects**

4 Short: Sense and Nonsense of Thermal Conductivity Enhancement in Liquid Chromatography

5

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12 **Declarations of interest:** none

13

14 **Abstract**

15 *We report on a numerical study of the thermal conductivity and temperature distribution in analytical*
16 *packed bed and monolithic HPLC columns to assess the feasibility of a number of potential solutions to the*
17 *viscous heating problem that would normally impede high efficiency separations when moving to extreme*
18 *operating pressures (e.g., 2500 bar). Computational fluid dynamic (CFD) simulations were employed to*
19 *study heat transfer on three hierarchical levels of the column: meso-pore level, through-pore level and*
20 *column level. At the first level, realistic values were determined for the conductivity of the porous zone*
21 *(k_{pz}), depending on the internal structure of the porous zone and the mobile phase used (acetonitrile, water*
22 *or a mixture of both). These k_{pz} -values were in turn used at the second level to determine realistic values*
23 *for the effective conductivity of the bed (k_{eff}). It was shown that the presence of a solid core only has a*
24 *minor effect on the packed bed conductivity. Using highly conducting materials as core material can be*
25 *expected to maximally lead to a 60% increase in bed conductivity. Contrarily, in monolithic beds, the*
26 *presence of a core material would form one continuous phase of highly conducting material, thus greatly*
27 *enhancing the conductivity of the bed. At the third level, the temperature field in the entire column (bed*
28 *and column housing) was resolved for three typical boundary conditions: isothermal, adiabatic and still-*
29 *air oven. The effect of different physical properties (inlet pressure, mobile phase composition, bed*
30 *conductivity, wall conductivity and column ID) on these temperature fields was investigated. It was shown*
31 *that, theoretically, besides 1mm ID columns also “core-shell monoliths” can provide a solution to viscous*
32 *heating (by increasing the bed conductivity). Other possible solutions are proposed and discussed.*

33 **Keywords:** viscous heating; thermal conductivity; Computational Fluid Dynamics; core-shell particles;
34 UHPLC; high pressure

35

36 1. Introduction

37 To decrease analysis time and improve separation performance, a reduction in particle size is the most
38 straightforward method in liquid chromatography. However, to operate sufficiently long columns packed
39 with these smaller particles (sub-2 μm), increased operating pressures are required. For a fixed column
40 length and flow rate, pressure drop increases with the inverse particle size squared ($1/d_p^2$). When working
41 at the optimal velocity, which increases with $1/d_p$, the pressure increase is even proportional to $1/d_p^3$. The
42 much larger operation pressure in current state-of-the-art ultra-high performance LC instruments (1200-
43 1500bar), requires to consider the thermal effects that result from pumping a liquid through a porous
44 medium. This so-called viscous heating or viscous dissipation of the mechanical energy increases the
45 temperature T of the mobile phase, column bed and hardware (wall, fittings, frits). The heat either exits
46 the column at its outlet (giving rise to axial temperature gradients) or through its wall (giving rise to radial
47 temperature gradients) ^[1-7]. In the limiting cases, all generated heat is either stored in the liquid (perfectly
48 adiabatic conditions, large axial temperature gradient), or removed through the wall (perfectly isothermal
49 column wall, large radial temperature gradient) ^[4,5,7]. Under practical LC conditions, where the column
50 hangs in a thermostatted compartment filled with air, the thermal boundary conditions will fall in between
51 these limiting cases, resulting in both axial and radial temperature gradients, with their amplitude
52 depending on the flow conditions around the column (still or forced air oven). For a fixed operating
53 pressure, the amplitude of the axial gradient depends mainly on the employed solvent (heat capacity),
54 whereas the radial gradients depend on the square of the column radius, the flow rate and effective
55 thermal conductivity k_{eff} of the column bed (particles + inter- and intra-particle mobile phase). The effects
56 of viscous heating can strongly be reduced by using capillary columns ^[8] however, in practice, standard
57 column formats (2.1-4.6mm) still account for the vast majority of the LC market ^[9]. A considerable amount
58 of work has already been done in literature concerning the potential solutions to this viscous heating
59 problem. However, most of these studies used a theoretical model (Zarichnyak) to estimate the thermal
60 conductivity of the packed bed. A recent computation fluid dynamics study from our group showed this
61 model is invalid for core-shell particles under the conditions prevailing in liquid chromatography ^[10]. These
62 simulations also showed that one of the previously proposed solutions, i.e. the use of core-shell particles
63 with a super-conducting core, cannot be expected to give more than a 50 to 60% increase of the thermal
64 bed conductivity, even when using core materials such as copper or gold which have a conductivity that
65 is orders of magnitudes larger than silica.

66 In the present study, we use the same computation fluid dynamics approach to assess the level of bed
67 conductivity enhancement that would be needed to keep the axial temperature increase and the radial
68 temperature gradients within affordable limits when contemplating the use of pressures exceeding the
69 currently available maximal commercial instrumentation pressure, i.e. in the range of 2000-2500 bar. In
70 addition, we also address the question whether the required enhancement is feasible or conceivable with
71 existing materials or column fabrication procedures.

72 For this purpose, a numerical heat transfer simulation study was conducted at three hierarchical levels:

- 73 i) meso-pore level
- 74 ii) through-pore level (=packing level)
- 75 iii) column level

76 The simulations at the meso-pore level are conducted to first determine the effective heat conductivity
77 (k_{pz}) of meso-porous silica, which is the material used in the vast majority of the currently employed
78 chromatographic media (particles, silica monolith). To represent the microscopic details of the silica
79 backbone making up the meso-porous materials, two general classes of non-porous solid phase structures
80 were considered: i) consolidated packings of contacting spheres and ii) sponge-like monolithic skeletons.
81 The first class represents the type of meso-porous materials obtained by agglomerating smaller nano-
82 spheres. The second group is used to represent the type of meso-porous materials obtained via spinodal
83 decomposition sol-gel processes. In the packed sphere case, a variety of different packing geometries is
84 possible (face-centered cubic (fcc); body-centered cubic (bcc); simple cubic (sc); random sphere packing,
85 see Figs. 1a-d). In each of these packing arrangements, the packing density (generally described using the
86 porosity ε) is fixed by the requirement that the particles need to be in contact with each other in order to
87 form a stable material. In the monolithic skeleton case, the porosity ε can, at least conceptually, be chosen
88 as an independent variable. To represent this type of geometry, we used the Tetrahedral Skeleton Model
89 (TSM) introduced by our group in ^[11], based upon a unit cell mimicking the tetrahedral bonding structure
90 in a diamond lattice. By varying the width and length of the skeleton branches, this geometry can
91 represent a very wide range of porosities, including the typical value of $\varepsilon=0.6$ to 0.65 of most underivatized
92 silica-gel materials. To broaden the range of investigated geometries, we also considered the case of the
93 co-called inverted TSM, wherein the solid and liquid regions of the standard TSM are swapped.

94 In the second part of the study, targeting the macroscopic packing level, the effective conductivities of
95 the meso-porous media obtained in the first part are combined with the effective conductivity of the
96 mobile phase mixtures typically employed in liquid chromatography to calculate the overall thermal
97 conductivity (k_{eff}) of the chromatographic bed. This is done in the absence as well as in the presence of an
98 interstitial flow. We also considered both fully porous and core-shell chromatographic media, as well as
99 both particulate beds and monolithic skeleton beds. In case of a monolithic bed, the “core shell” case
100 consists of a fully solid skeleton (the core) covered uniformly with a layer of meso-porous material.

101 In the third part of the study, the k_{eff} -values obtained in part two are used in a whole-column model to
102 calculate the steady-state radial and axial thermal gradients one can expect in a typical metal-tube based
103 commercial chromatographic column for UHPLC (dimensions 2.1x50mm) for the three “classic” thermal
104 boundary conditions (isothermal, adiabatic and still-air oven) and for varying degrees of the thermal
105 conductivity of the packing material. This allows to quantify which magnitude of thermal conductivity (and
106 hence also which type of materials) would be needed to keep the thermal gradients within affordable

107 limits when considering running columns at pressures in the 2000 to 2500 bar range. Next to an increase
108 of the bed conductivity, a number of other potential solutions to alleviate the viscous heating temperature
109 gradients are discussed as well.

110 **2. Numerical methods**

111

112 **2.1 Geometries and grid**

113 Fig. 1 shows all geometries used for the conductivity simulations at the meso-pore and through-pore level.
114 The column geometry (including detailed dimensions) used in the third part of the study can be readily
115 viewed from temperature contour plots shown in Fig. 7 further on. These dimensions were determined
116 by measuring a number of widely used commercial columns and rounding the resulting average value.
117 Furthermore, some small simplifications of the actual design (e.g. the assumption of cylindrical symmetry
118 for the column end-fitting pieces) were made as a compromise between required computational effort
119 and gain in extra information.

120 In all considered spherical core-shell materials, a relative core radius $\rho=0.63$ was used. This yielded a
121 volume fraction of shell material of 0.75. For the tetrahedral skeleton a relative core radius $\rho=0.45$ was
122 used. This yields the same volume fraction of shell material of 0.75 in the $\varepsilon_e=0.60$ case and a slightly
123 different fraction (0.73) in the $\varepsilon_e=0.39$ case.

124 On the meso-pore and packing level, all ordered sphere packings contained at least 8.5×10^5 cells. All TSM
125 geometries contained at least 1.9×10^6 cells. Both types of geometries were meshed with mainly
126 tetrahedral cells, and inflation was used to ensure smaller, triangular prism shaped cells near the
127 core/shell interface and shell/liquid interface as here the highest velocity and temperature gradients
128 occur. The sizing of the mesh cells was such that quadrupling the number of cells had an impact of less
129 than 0.1% on the measured effective conductivity.

130 “The random packing contained 3.2×10^6 tetrahedral cells, resulting in 2.4×10^3 cells per particle. Cells
131 sizes were smallest near the interfaces. As a grid check, the SC packing was meshed with the same settings
132 as the random packing (also yielding 2.4×10^3 cells per particle) and used with some typical material
133 conductivities, resulting in a maximal error^[10] on the effective conductivity of 1.2%.”

134 On the column level, all geometries were meshed with at least 220.000 rectangular cells. In the packed
135 bed, smaller cells were used (as compared to the column wall and end fittings) to accurately calculate the
136 velocity profile and cells were smaller in the radial direction (as compared to the axial direction), because
137 in this direction velocity and temperature gradients are higher. Inflation was used to ensure smaller cell
138 sizes near the bed/column wall interface, because in this region the highest velocity and temperature
139 gradients are present. Halving the cell size in both axial and radial direction led to a change in the reported
140 temperature differences ($\Delta T_{\text{rad,bed}}$, $\Delta T_{\text{ax,bed}}$, $\Delta T_{\text{ax,col}}$, see later) of less than 0.2%.

141

142 **2.2 Simulation procedures**

143 Simulations of the effective conductivity at the meso-pore and through-pore level were carried out using
144 the same steady-state method described and accurately validated in ^[10]. In brief, the energy equation
145 (which describes the conservation of energy principle) ^[12] was solved using the finite volume solvers of
146 Ansys Fluent to find the steady-state temperature field resulting from two opposite boundaries at
147 constant, but different temperature. This allows reporting the heat flux through the geometry. The
148 analytical solution (derived from Fourier's heat transfer law) describing the heat flux through an infinitely
149 wide slab of finite thickness and consisting of a homogeneous material with thermal conductivity k_{eff} was
150 used to find the value of k_{eff} that correctly predicts this reported heat flux.

151 Simulations of the temperature distribution at the column level were carried out under steady-state
152 conditions as well. Conservation equations for mass and momentum (to solve the velocity field) were
153 solved together with the equation for energy conservation (to solve the temperature field). All equations
154 were solved simultaneously because the velocity and temperature field are coupled (through viscous
155 heating and through the temperature dependency of density and viscosity).

156 Liquid viscosity was put at Newtonian. The velocity fields were obtained by solving the Navier-Stokes
157 equations using the coupled pressure-based steady state solver with a Least-Squares Cell-Based gradient
158 evaluation and a second order upwind interpolation scheme for the momentum and second order for
159 pressure. All velocity profiles were calculated by imposing a fixed pressure at the inlet and outlet.
160 Temperature fields were obtained using a second-order upwind spatial discretization. Spatial gradients
161 were again evaluated using the Least-Squares Cell-Based method. The meshing density was typically on
162 the order of 800 cells/mm².

163 Pressure inlet and pressure outlet boundary conditions were used, with a range of different inlet
164 pressures, while the outlet pressure was fixed at 1 atm. The mobile phase inlet temperature was 298.15
165 K. The base of the column geometry was assigned an axisymmetric boundary condition. This makes that
166 this 2D geometry behaves as a 3D geometry with cylindrical symmetry. At the interface between the fluid
167 zone and capillary or column wall (subsequently in the inlet capillary, cone, frit, packed bed, frit, cone and
168 outlet capillary) a no slip boundary condition was used. All capillary and column walls were assigned the
169 properties of stainless steel ($k=16.3 \text{ W(m.K)}$). At the outer boundaries of the capillaries and column wall
170 three different thermal boundary conditions were used: adiabatic boundary conditions to mimic a
171 perfectly insulated column, isothermal boundary conditions ($T_{\text{wall}}=298.15 \text{ K}$) to mimic a perfectly
172 thermostatted column and natural convection boundary conditions ($T_{\text{room}}=298.15 \text{ K}$) to mimic a still-air
173 oven. Natural convection coefficients were calculated according to Eq. 10 from Churchill and Chu (1975)
174 ^[13].

175 Three different liquids were used as mobile phase: pure water (H₂O), pure acetonitrile (ACN) and a
 176 40/60(v%/v%) ACN/H₂O mixture. Density and viscosity of these liquids as a function of temperature and
 177 pressure were taken from National Institute of Standards and Technology (NIST) database in case of water
 178 and from Billen et al. ^[14] in case of ACN and the ACN/H₂O mixture. Data was fitted to provide the solver
 179 with a continuous function of pressure and temperature. In ^[14] data is only reported for pressures up to
 180 1000 bar, hence for higher pressures these data were extrapolated. The expansion coefficient $\alpha(p,T)$ was
 181 calculated from these fitted functions as:

$$182 \quad \alpha(p, T) = \frac{1}{\rho} \frac{\partial \rho}{\partial T} \quad (1)$$

183 wherein ρ is the density of the liquid. Thermal conductivity (k) and isobaric heat capacity (C_p) were taken
 184 at constant p (1 bar) and T (298.15 K).

Table 1. Mobile phase thermal properties

Mobile phase liquid	k [W/(m.K)]	C _p [J/(kg.K)]
H ₂ O	0.61 ⁽¹⁾	4181.3 ⁽¹⁾
ACN	0.19 ⁽²⁾	2233.6 ⁽³⁾
40/60(v%/v%) ACN/H ₂ O	0.41 ⁽⁴⁾	3785.8 ⁽³⁾

(1) NIST

(2) Hulse et al., 2004. ^[15]

(3) Kolker and Safonova, 2010 ^[16]. In case of the mixture, the C_p value was determined by linear interpolation between the two closest mixture compositions.

(4) By interpolation between pure H₂O and ACN, according to Jamieson et al. ^[17] with $\alpha=1$.

185

186 While the mobile phase is ‘pushed’ through the system, energy originating from the effected pressure
 187 work is dissipated in the system. Part of it is absorbed during the expansion of the mobile phase liquid,
 188 while the remainder leads to the heating of the mobile phase liquid. Eq. (2) gives the pressure work done
 189 per unit of time per unit of volume (W) in an infinitesimal control volume in cylindrical coordinates.

$$190 \quad W = \left(\frac{\partial(u_x P)}{\partial x} + \frac{1}{r} \frac{\partial(r u_r P)}{\partial r} \right) \quad (2)$$

191

$$192 \quad = \left(u_x \frac{\partial P}{\partial x} + P \frac{\partial u_x}{\partial x} + P \frac{\partial u_r}{\partial r} + u_r \frac{\partial P}{\partial r} + \frac{P u_r}{r} \right)$$

193

194 In which P is the local pressure and u_x and u_r the axial and radial velocity components.

195 This pressure work term multiplied with αT , in which α is the thermal expansion coefficient, gives the
 196 amount of energy absorbed by the thermal expansion and hence the remaining energy, available for
 197 heating is ^[7,18]:

$$\left(u_x \frac{\partial P}{\partial x} + P \frac{\partial u_x}{\partial x} + P \frac{\partial u_r}{\partial r} + u_r \frac{\partial P}{\partial r} + \frac{P u_r}{r} \right) (1 - \alpha T)$$

198

(3)

199

200 A user defined function (UDF), that allows to incorporate this expression in the energy balance as an extra
201 energy source term was used.

202 All simulations were performed with the finite volume solvers of Ansys Fluent, version 17.1 from Ansys,
203 Inc. This software was used on Dell Power Edge R210 RackServers each equipped with an Intel Xeon x3460
204 processor (clockspeed 2,8 GHz, 4 cores) and 16 Gb, 1333 MHz ram memory and Windows server edition
205 2008 R2 (64-bit) as operating system.

206 **3. Results and discussion**

207 ***3.1 Simulations at the meso-pore level***

208 In this section, heat conductivity simulations were carried out at the meso-pore level, considering binary
209 media consisting of an impermeable silica network (the silica backbone) whose interstices are filled with
210 a liquid medium (the mobile phase zone) as depicted in Fig. 1. The obtained conductivity values represent
211 the effective conductivity k_{pz} of the meso-porous zone in silica materials when filled with the mobile phase
212 liquids typically used in LC. Given the thermal conductivity of alkyl chains/ C_{18} ($k_{C18}=0.15W/(m.K)$) is not
213 too different from that of the typical mobile phases being used in chromatography (typically ranging
214 between $k_m=0.61 W/(m.K)$ for pure water and $0.19 W/(m.K)$ for pure ACN), and given that in most cases
215 the alkyl chains anyhow make up only a fraction of the stationary phase layer, the latter is assumed to be
216 an integral part of the mobile phase zone.

217 Fig. 2 shows how the obtained effective conductivity curves monotonically decrease with increasing ε .
218 This increase can be understood as a direct consequence of the fact that an increase in ε inherently implies
219 a decrease of the fraction of silica and the fact that silica has a markedly larger conductivity ($k_{SiO_2}=1.40$
220 $W/(m.K)$) than the mobile phase.

221 It can also be noted that the effective conductivity curves shift upwards when the conductivity of the
222 mobile phase (k_m) increases. This increase is less than linearly proportional because the mobile phase only
223 makes up part of the material. For the same reason (effect of k_m is damped by presence of silica material),
224 it is obvious to observe the effect of k_m is smallest in the structures with the lowest external porosity ε
225 and gradually becomes larger when ε increases, i.e., with increasing fraction of mobile phase.

226 Another observation from Fig. 2 is that the geometrical details of the structure have a smaller effect than
227 the porosity. For the TSM and inverted TSM-models, this even holds to such an extent that both structures
228 produce nearly perfectly overlapping curves (cf. black solid and open symbols). The difference between
229 the tetrahedral skeleton structures (connected black data points) and the sphere packings (isolated blue
230 and red data points) is also relatively small, reducing to virtually nothing when the mobile phase

231 conductivity is large (blue data points nearly coincide perfectly with the black data point curves for the
232 case of $k_m=0.58$ W/(m.K)). Not surprisingly, the $k_m=0.58$ -case is the one where the geometrical details
233 matter least, as this is also the case where the difference in conductivity between the solid and liquid
234 phase is smallest. When the mobile phase conductivity is small (cf. position of red data points with respect
235 to black data points for $k_m=0.21$ W/(m.K)), the particulate bed geometry produces effective conductivity
236 values that are about 10-15% lower than the tetrahedral skeleton. This effect can be owed to the higher
237 degree of connectivity in the tetrahedral structure as opposed to the limited contact area between the
238 individual particles in the consolidated particle packing.

239 **3.2 Simulations at the packing and through-pore level**

240 Moving up one level, the present section considers the effective heat transfer occurring at the level of the
241 interstitial space and the particles (or skeleton branches in case of monolithic media). The overall
242 conductivity calculated at this level is hence representative for the effective conductivity k_{eff} of the
243 chromatographic bed. In these calculations, the k_{pz} -values obtained in the previous section are used to
244 represent the thermal conductivity of the meso-porous zones of the particles or the skeleton branches.
245 For the thermal conductivity of the mobile phase liquid in the interstitial space, the well-established values
246 for water and acetonitrile are used. Both the pure liquids as well as a 40/60(v%/v%) ACN/H₂O mixture
247 were considered. For the sake of brevity, most presented results are for water.

248 First, the effect of the presence of a convective flow in the interstitial space was investigated. As can be
249 noted from Fig. 3, the effective bed conductivity of a 2 μ m particle bed can only be expected to be
250 significantly influenced by the presence of a convective heat transfer component when the fluid velocity
251 exceeds 10 cm/s. These are values that are normally never reached in chromatography. All subsequent
252 calculations have therefore been conducted assuming a zero velocity in the through-pores, in agreement
253 with the assumptions made in earlier work ^[19].

254 As already stated, Fig. 2 shows that the k_{pz} -value of a typical silica sol-gel material (for which the internal
255 porosity before coating is around $\varepsilon=0.6$ to 0.65) is typically of the order of $k_{pz}=0.5$ to 0.9 W/(m.K).
256 However, to cover the widest possible range of k_{pz} -values, the simulations performed in Figs. 4-5 were
257 carried out over a range that is significantly broader ($0.2 < k_{pz} < 2$) than is accessible with silica as the
258 support material to offer a more general view on the relation between the k_{eff} of the bed and the k_{pz} of
259 the mesoporous silica material. Fig. 4 compares the effective bed conductivity k_{eff} for two sphere packing
260 cases: a perfectly ordered one (fcc packing with a porosity of $\varepsilon=0.24$, Fig. 4a) and a randomly packed one
261 (Fig. 4b) with a porosity of $\varepsilon=0.39$ as typically encountered in packed bed columns. Fig. 5 shows the k_{eff} -
262 data for two distinct tetrahedral skeleton model cases, resp. with an external porosity of $\varepsilon=0.4$ and 0.6.
263 The first value was selected to enable a direct comparison with the packed bed of spheres, while the
264 second value is more typical of actual silica monolithic beds ^[20].

265 Considering first only the black curves (=no core) in both Fig. 4 and 5, it can be observed how k_{eff} increases
266 with increasing conductivity (k_{pz}) of the packing material, as physically expected. Again, this increase is
267 weaker than linear because the meso-porous material, i.e., the region where the variation in k_{pz} is actually
268 imposed upon, only makes up a fraction ε of the total bed, while the conductivity of the mobile zone
269 (remaining fraction $1-\varepsilon$) is kept invariant at $k_m=0.58$. For similar reasons, it is also obvious to see that the
270 increase of k_{eff} with k_{pz} is steepest for the material with the highest solid fraction (fcc with $\varepsilon=0.24$, fig. 4a)
271 and weakest for the material with the lowest solid fraction (TSM with $\varepsilon=0.6$, fig. 5b).

272 On a side note, it is also interesting to observe that, in the absence of a solid core (=black data in Figs. 4
273 and 5), the porosity effect is significantly larger than the effect of the packing shape, as can for example
274 be witnessed from the fact that the $\varepsilon=0.39$ -packed bed and the $\varepsilon=0.39$ -tetrahedral skeleton model (resp.
275 Fig. 4b and Fig. 5a) basically lead to identical k_{eff} -curves, while structures of the same type but with
276 different ε lead to significantly different k_{eff} -values (compare Fig. 4a with 4b or compare Fig. 5a with 5b).

277 The most striking observation from Figs. 4 and 5 is made when considering the data points corresponding
278 to the presence of a non-porous, high conductivity core (=gray symbols and curves). In ^[10] it was already
279 observed and explained that the use of particles with high-conductivity cores can only be expected to
280 have a limited effect on the overall bed k_{eff} . This is confirmed again in Fig. 4, showing that, for example,
281 the presence of a solid silica core only changes the effective bed conductivity over that of a bed of fully
282 porous particles with some 10% to 15% in the practically relevant range of k_{pz} -values (ranging from 0.8 to
283 1.1 W/(m.K) in case of water as a mobile). Even when contemplating the use of Al_2O_3 - or Cu-cores, with a
284 conductivity that is respectively 29 and 285 times higher than that of a silica core, the increase in effective
285 over-all bed conductivity is nowhere larger than some 50 to 60%.

286 Whereas Fig. 4 is for pure water, we found it also instructive to investigate the changes in k_{eff} with varying
287 core conductivity for pure ACN as well as for a mixture with some intermediate composition, somewhat
288 arbitrarily chosen at 40%/60%(v/v) ACN/ H_2O , as this leads to k_{eff} -values situated somewhere midway the
289 k_{eff} of pure water and that of pure ACN. The results are shown in Fig. 6, again confirming maximal gains
290 can in all three cases never be expected to become larger than some 55%. The course of the three curves
291 is nearly identical (relative plots nearly perfectly overlap, data not shown). The vertical difference between
292 the curves can hence be fully attributed to the differences in conductivity of the mobile phase, with the
293 highest conductivity mobile phase (pure water) producing the highest k_{eff} , as physically expected.

294 On the other hand, the use of a high conductivity core material clearly would have a much stronger effect
295 in the monolithic bed case (Fig. 5), as the computed k_{eff} -values are roughly one order of magnitude larger
296 than in the core-shell particle bed cases (compare y-axis scale with Fig. 4). The straightforward explanation
297 for this difference is that in the spherical particle case the high conductivity cores are isolated from each
298 other by the less conductive shells and can hence not form an interconnected high conductivity path
299 through the column. In the monolithic case, on the other hand, the core backbone inherently forms a fully

300 interconnected structure, such that any higher conductivity automatically carries over the entire length
301 of the bed.

302 Comparing the $\varepsilon=0.39$ and $\varepsilon=0.6$ -monolithic structures (Figs. 5a-b) shows the $\varepsilon=0.39$ -case leads to higher
303 k_{eff} -values than the $\varepsilon=0.6$ -case. This can again be explained by the higher fraction of highly conductive
304 solid material in the $\varepsilon=0.39$ -case.

305 **3.3 Simulations at the column level**

306 Finally, moving to the whole-column level to address the main research question of our study, the k_{eff}
307 values calculated in the preceding sections are used to calculate the temperature distribution profiles that
308 can be expected in conventional metal tubing-based 5cm columns operated at pressures up to 2500bar.
309 Fig. 7 shows the results obtained at 2500bar for the three different considered thermal boundary
310 conditions. In general, the temperature contour plots are very similar to the ones already obtained in
311 literature at lower pressures [6,7,19,21-23]. They display similar shape, but higher absolute temperature
312 values. This implies that the main effect of the higher inlet pressures employed is due to the accompanying
313 increased viscous heating, while secondary effects resulting from the high temperatures and pressures
314 (density and viscosity changes) do not lead to a specific deviation from the thermal behavior observed at
315 lower pressures.

316
317 Similar to what is already well-documented for lower operating pressures [6,7,19,21-23], the isothermal lines
318 in a column operated at 2500 bar still substantially run in the axial direction when the column is operated
319 in a thermostatted environment (Fig. 7a). Some small deviations from a perfect axial uniformity can be
320 seen (mainly at the inlet of the bed), but these can essentially be attributed to entrance effects and the
321 presence of the endfittings and the metal inlet tubing, details that were not included in earlier work. In
322 the adiabatic case on the other hand, the isothermal lines substantially run in the radial direction (Fig. 7b),
323 reflecting that here the temperature mainly changes in the axial direction, again in agreement with the
324 situation at lower pressures. The small radial gradient that can be observed despite the adiabatic
325 conditions is caused by the fact that the adiabatic wall conditions are imposed at the outer mantle of the
326 metal tubing, leaving the possibility for an internal heat flow running from the end to the front of the
327 column along the metal tubing [7].

328
329 Fig. 7c shows that the overall temperature distribution under natural convection conditions (representing
330 the practical conditions in a still-air oven) lies much closer to the adiabatic case than to the isothermal
331 case. Since the temperature difference between the inside of the column and the environment is smaller
332 at the start of the column, the radial temperature gradients are the smallest at the inlet and increase
333 towards the end of the column. This is again similar to the situation observed at lower pressures [6,7,19,21-
334 23].

335

336 3.3.1 Column-level temperature gradients

337 Comparing the results of all conducted whole-column simulations in a more quantitative way, Fig. 8 shows
 338 a plot of the axially-averaged radial temperature difference ($\Delta T_{rad,bed}$, see Eq. (5)) measured across the
 339 packed bed part of the column as a function of the radially-averaged axial temperature difference ($\Delta T_{ax,bed}$,
 340 see Eq. (4)) for different inlet pressures for the case of a 40/60 (v/v) ACN/water mixture (very similar
 341 curves are obtained for pure water and pure ACN-cases, see Fig. 1 and 2 of SM). As a side note, it should
 342 be remarked that, when changing the pressures, it was opted to keep the flow rate constant at 0.5 mL/min
 343 (measured at column inlet). With also the column length fixed, this implies gradually decreasing column
 344 permeabilities (corresponding to ever decreasing particle sizes) are considered when the pressure was
 345 ranged from 500 to 2500bar. Under the presently adopted conditions of a fixed bed length $L=5m$ and flow
 346 rate $F=0.5mL/min$, the particle sizes corresponding to these different pressures respectively correspond to
 347 $1.53\mu m$ at 500bar, $1.22\mu m$ at 750bar, $1.04\mu m$ at 1000bar, $0.82\mu m$ at 1500bar, $0.69\mu m$ at 2000bar and
 348 $0.59\mu m$ at 2500bar. Other assumptions are certainly possible, but it was considered that keeping the
 349 permeability constant would lead to flow rates that would become impractically large when reaching the
 350 highest pressures. Higher pressures can obviously also be used to make longer columns, but in that case
 351 the viscous heating problem is less important (F decreases with increasing L and consequently the amount
 352 of produced heat $Q=F \cdot \Delta P$ decreases).

353
 354 In Fig. 8a, $\Delta T_{rad,bed}$ is plotted versus the axial temperature difference between the column in- and outlet
 355 ($\Delta T_{ax,col}$). This is the axial temperature increase measure that is experimentally most easily obtained (by
 356 inserting a thermocouple in the in- and outlet stream of the column) and therefore used in many studies
 357 on viscous heating. In Fig. 8b, the x-axis coordinate represents the axial temperature establishing over the
 358 packed segment of the column only ($\Delta T_{ax,bed}$). This is the chromatographically more relevant measure, and
 359 will hence be used in the remainder of the study. Since the temperature in the bed varies with both the
 360 axial and radial position, it should be noted that the reported $\Delta T_{ax,bed}$ was calculated by taking the
 361 difference between the average (mixing cup) temperature at the bed inlet and outlet and $\Delta T_{rad,bed}$ by
 362 taking the average along the total bed length of the temperature difference between the center and outer
 363 region of the bed.

$$364 \Delta T_{ax,bed} = \bar{T}_{bed,out} - \bar{T}_{bed,in} = \frac{\int_{S_{bed,out}} \rho \cdot C_p \cdot T \cdot u \cdot \varepsilon \cdot dS}{\int_{S_{bed,out}} \rho \cdot C_p \cdot u \cdot \varepsilon \cdot dS} - \frac{\int_{S_{bed,in}} \rho \cdot C_p \cdot T \cdot u \cdot \varepsilon \cdot dS}{\int_{S_{bed,in}} \rho \cdot C_p \cdot u \cdot \varepsilon \cdot dS} \quad (4)$$

$$368 \Delta T_{rad,bed} = \frac{\int_0^L (T_{axis} - T_{wall}) \cdot dx}{L} \quad (5)$$

370

371 With, \bar{T} the mixing cup temperature, ρ the density, C_p the heat capacity, u the x-velocity, ε the bed
372 porosity, S the cross sectional area of the bed, L the length of the bed, T_{axis} the temperature at the axis of
373 the bed and T_{wall} the temperature at the interface between bed and column wall.

374
375 First considering the plots of $\Delta T_{\text{rad,bed}}$ versus $\Delta T_{\text{ax,col}}$ (Fig. 8a), the isothermal boundary condition case
376 clearly leads to the physically expected behavior: no significant change in temperature along the x-axis,
377 while the radial temperature difference across the bed consistently increases with increasing pressure.
378 Hence the near-perfectly vertical progression of the data points. The adiabatic boundary condition on the
379 other hand produces the strongest axial temperature increase while the radial T-gradient is smallest.
380 However, and despite what is physically expected for an adiabatically insulated system, the $\Delta T_{\text{rad,bed}}$ -values
381 are not negligible and steadily increase with increasing pressure. This is due to the already mentioned
382 heat backflow in the metal column mantle. This backflow establishes because in practice the adiabatic
383 wall condition can only be applied on the outer wall of the metal tubing, such that part of the heat can
384 still radially escape from the chromatographic bed region. The natural convection case shows a behavior
385 that is very similar to the adiabatic case, but with a $\Delta T_{\text{rad,bed}}$ versus $\Delta T_{\text{ax,col}}$ -relationship that has a roughly
386 two-fold higher slope. In both cases (natural convection and adiabatic), the relationship between $\Delta T_{\text{rad,bed}}$
387 and $\Delta T_{\text{ax,col}}$ is close to linear.

388
389 Comparing Fig. 8a with Fig. 8b, where the x-axis represents $\Delta T_{\text{ax,bed}}$ instead of $\Delta T_{\text{ax,col}}$, it readily becomes
390 apparent that the axial temperature differences across the bed ($\Delta T_{\text{ax,bed}}$) are considerably smaller than
391 those measured across the entire column, with roughly $\Delta T_{\text{ax,bed}} \cong 60\%$ of $\Delta T_{\text{ax,col}}$. Or in other words, about
392 40% of the measured temperature increase occurs in the metal endfitting pieces. This shows the reduction
393 in retention caused by the viscous heating is significantly less than one would expect when determining
394 the axial temperature increase by inserting thermocouples in the in- and outlet flow. Taking the case
395 represented in Fig. 8 (40/60 ACN/H₂O mixture, $F_{\text{in}}=0.5\text{mL/min}$, $L=5\text{cm}$), the axial temperature across the
396 bed itself would be limited to some 18°C at 2500bar under still air conditions (=natural convection), rather
397 than the 32°C measured across the column. Taking a typical retention enthalpy of $7.5 \cdot 10^3 \text{ J/mol}$ this
398 corresponds to a decrease in retention between front and end of the bed of some 15%, whereas the
399 across-column temperature difference would one make to assume a 25% loss in retention capacity
400 between in- and outlet. Although the course of the curves in Fig. 8b is very similar to those in Fig. 8a, they
401 show a distinct upward curvature, as the values on the y-axis are the same as in Fig. 8a, where $\Delta T_{\text{ax,col}}$
402 increases almost linearly with ΔP , but $\Delta T_{\text{ax,bed}}$ increases slower (due to an increasing fraction of the
403 removed heat being lost in the large surface area column endfittings). Plotted versus $\Delta T_{\text{ax,bed}}$, the
404 isothermal case now also displays a small axial temperature difference. This temperature increase is
405 eliminated in $\Delta T_{\text{ax,col}}$ as this heat is lost in the narrow outlet tubing due to conduction to the column
406 endfitting.

407
408 Whereas the data shown in Fig. 8 were for a 40/60 ACN/water mixture, it should be noted the slope of
409 the $\Delta T_{rad,bed}$ versus $\Delta T_{ax,bed}$ -curves strongly depends on the composition of the mobile phase. This is shown
410 in Fig. 9, where the mixture data already shown in Fig. 8 are compared to those obtained for pure water
411 and pure ACN. This is done for both the natural convection (Fig. 9a) and the adiabatic thermal boundary
412 condition (Fig. 9b). In both cases, there is a considerable increase of the slope of the $\Delta T_{rad,bed}$ versus
413 $\Delta T_{ax,bed}$ -curves when going from the purely aqueous to the purely ACN mobile phase, with the 40/60
414 ACN/water mixture lying in between both extremes. The increase in slope with increasing ACN-fraction
415 can be understood from the decrease in radial conductivity of the packed bed due to the much lower
416 thermal conductivity of ACN compared to water. The increase in $\Delta T_{ax,bed}$ is on the other hand related to
417 the combined effect of the difference in mobile phase heat capacity, the amount of generated heat
418 (proportional to flow rate and pressure drop) and heat lost to the environment. In the adiabatic case (Fig.
419 9b), the $\Delta T_{rad,bed}$ -values are clearly smaller than in the natural convection case, but remain very significant
420 and display a similar relative dependency on the mobile phase composition. The data also show that
421 developing solutions to create the perfect adiabatic environment for the column are not sufficient if
422 nothing is changed to lower the backflow of heat along the metal column wall.

423
424 Please note that the change in mobile phase considered in Fig. 9 is made under the assumption the
425 permeability of the bed at a given pressure is kept identical to that of the ACN/water-mixture. This implies
426 the flow rates in the pure water and the pure ACN-case no longer correspond to $F=0.5\text{mL/min}$ but are
427 changed inversely proportional with the change in viscosity, yielding a 1.56-2.28 times higher flow rate in
428 the ACN case and a 1.06-1.17 times smaller flow rate in the H_2O case (depending on the inlet pressure and
429 thermal boundary conditions). The results obtained for the case where F_{in} is kept constant for every
430 considered liquid composition and every pressure is shown in the SM (Fig. S3). As can be noted the effects
431 are qualitatively similar, but less outspoken as cases with the same inlet pressure now also represent a
432 same amount of heat generated.

433
434 While the pressure coordinate in Figs 8 and 9 has been eliminated by directly plotting $\Delta T_{rad,bed}$ versus
435 $\Delta T_{ax,bed}$, the corresponding plots of $\Delta T_{rad,bed}$ versus ΔP and $\Delta T_{ax,bed}$ versus ΔP are respectively shown in Figs.
436 10 and S4 of the Supplementary Material (SM). Fig. 10 shows $\Delta T_{rad,bed}$ increases linearly with increasing
437 ΔP up to the highest considered ΔP (Pearson's correlation coefficient >0.999), with the slope depending
438 on the mobile phase composition and thermal boundary conditions.

439
440 The nearly linear relationship with ΔP allows to conclude that moving to a 2500bar operation would simply
441 double the radial temperature gradient compared to working at a currently typical maximal pressure of
442 1250bar. Fig. 9 shows that at 1250bar $\Delta T_{rad,bed}$ in a 2.1 mm column can be expected to be of the order of

443 0.75°C for pure water, some 1.5°C for the ACN/water mixture and some 2.5°C for pure ACN. The loss in
444 column efficiency these temperature differences are generating is something the chromatographic
445 community has, out of necessity, learned to live with. Considering now that a doubling of the inlet
446 pressure would double the radial temperature differences, it follows immediately that, in order to bring
447 these differences back to the current level, a solution needs to be found that could halve the difference
448 again.

449

450 3.3.2 Bed conductivity enhancement solutions

451 In this and below section, it is investigated what solutions could exist to halve the radial temperature
452 gradients in a 2.1x50mm column operated at 2500 bar, or even reduce it below that level.

453

454 A solution already suggested in literature would consist of using particles with an enhanced thermal
455 conductivity ^[24]. Grinias *et al.* showed experimentally that the presence of a “regular” solid silica core
456 impacts the thermal behavior and efficiency of chromatographic columns ^[25]. To investigate this, Fig. 11
457 shows how the $\Delta T_{\text{rad,bed}}$ versus $\Delta T_{\text{ax,bed}}$ -curves can be expected to vary if the thermal conductivity of the
458 bed could be increased. Data are shown for the 40/60 ACN/water-mixture (Fig. 11a) as well as for pure
459 ACN (Fig. 11b). The pure water case is not considered, as the radial thermal gradient problem is less
460 stringent. The observed effect is furthermore very similar to that of the two other considered fluids.

461

462 The first important conclusion from Fig. 11 is that, within the investigated range of considered k_{eff} -
463 increases, the increase in k_{eff} has nearly no effect on the axial temperature difference, while the radial
464 temperature difference decreases following a nearly perfect inversely proportional relationship with k_{eff} :

$$465 \quad \Delta T_{\text{rad,bed}} \sim 1/k_{\text{eff}} \quad (6)$$

466 A second important conclusion is that roughly a doubling of k_{eff} would be needed in order to bring the
467 $\Delta T_{\text{rad,bed}}$ at 2500bar below its current value experienced at 1250bar (levels indicated by dashed horizontal
468 lines). As can be noted from Fig. 6, this is not possible with particulate beds (maximal enhancement=55%),
469 while such a doubling would be easily within reach if a monolithic bed structure consisting of a highly
470 conductive core could be produced. As can be seen from Fig. 5, the thermal conductivity enhancement of
471 monolithic structures can easily amount up to even a factor of 40, due to the strong degree of axial
472 connectivity of the monolithic structure.

473 3.3.3 Other solutions

474 Next to enhancing the radial heat transport, another way to suppress the creation of radial gradients, at
475 least conceptually, would be to replace the highly conductive metal of the column wall with a less
476 conductive material to stop the radial heat losses at the column inner wall instead of at its outer wall.
477 Provided such a material possessing the required mechanical properties can be found, this would

478 eliminate the aforementioned problem of the backflow of heat leading to the relatively large radial
479 gradients under perfect adiabatic conditions observed in Fig. 8 and 9b.

480
481 The effect of the conductivity of the column tube wall is investigated in Fig. 12, where the natural
482 convection and adiabatic case data for the 40/60 ACN/H₂O mixture obtained in a metal tubing column
483 already shown in Fig. 11a (solid red line in Fig. 12) are compared with the temperature gradient values
484 one can expect when the entire column (wall + endfitting pieces) would be made in a material that is
485 approximately 4, 16, 64 and 256 times less conductive than what is typical for stainless steel.

486
487 In agreement with one's physical expectations, making the column wall less conductive indeed leads to a
488 reduction of the radial temperature gradient. Since in steady-state, the produced heat needs to leave the
489 column anyhow, the reduction of the radial gradient is inevitably also accompanied by an increase of the
490 axial temperature gradient. From a practical point of view, the most interesting data set is the one for
491 $k_{\text{wall}}=0.25\text{W/m.K}$ as this is representative of a column that would consist entirely of PEEK. As can be noted,
492 such a column would suffer much less from radial temperature differences than a metal column
493 ($\Delta T_{\text{rad,bed}}=0.66\text{ }^{\circ}\text{C}$ instead of $2.13\text{ }^{\circ}\text{C}$ for a stainless steel wall column operated at 2500bar under natural
494 convection conditions). For the case of a 4-fold wall conductivity reduction (certain metal oxides such as
495 Al and Zr-oxide or rock like materials such as quartz, marble and granite have thermal conductivities in
496 this range), the effect on $\Delta T_{\text{rad,bed}}$ is still limited. It is only if a material with a 16-fold reduction
497 ($k_{\text{wall}}=1\text{W/m.K}$, representative of traditional low thermal conductivity ceramic materials ^[26]) is being used
498 that $\Delta T_{\text{rad,bed}}$ could be brought under 1°C (in case of natural convection).

499
500 Whereas increasing the thermal conductivity of the bed has barely any effect on the axial temperature
501 gradient ($\Delta T_{\text{ax,bed}}$, see Fig. 11), the use of low-conductivity column walls clearly leads to a significant
502 increase of the axial temperature increase. This is clearly a drawback of this approach.

503
504 The adiabatic-case data in Fig. 12b display a similar pattern, albeit the use of the lower conductivity
505 materials now really drives $\Delta T_{\text{rad,bed}}$ to values $<0.1\text{K}$, showing that if a change to a less conductive column
506 wall material could be combined with a perfect adiabatic operation (as would be obtained when surrounding
507 the columns with a vacuum jacket as proposed by Gritti ^[27]), the radial temperature gradients would be
508 reduced to a level where its effect on column band broadening would be truly insignificant. The increase
509 of the axial temperature gradient is however more pronounced than in the natural convection case shown
510 in Fig. 12a.

511
512 Since an integral PEEK column cannot be expected to withstand high internal pressures without
513 deformation, except when conceiving impractically thick column walls, we also found it instructive to
514 investigate whether it would suffice to coat only the inner wall of a metal tube with a thin ($100\text{ }\mu\text{m}$) PEEK

515 layer (keeping the column ID constant), using the PEEK layer for thermal insulation and the surrounding
516 metal tube for mechanical strength. As can be noted from the position of point A in Fig. 12, the effect of
517 the thin PEEK layer is relatively small and only lowers $\Delta T_{\text{rad,bed}}$ with some 17% (compared to the 69%
518 decrease of an integral PEEK column. This is due to the fact that the heat transfer resistance of the thin
519 layer is insufficient to block a sufficient part of the radial heat transfer, such that there remains a
520 significant backflow of heat along the metal part of the wall. In an integral PEEK column, not only the
521 radial heat transfer is blocked to a much more significant extent (column wall is about 20 times thicker
522 than 100 μm PEEK layer roughly leading to a 12 times larger thermal resistance), but also the axial back
523 flow of heat is suppressed to a much larger extent (conductivity of PEEK is about 64 times less than that
524 of stainless steel). As can be noted from interpolating between points C and D, a 100 μm thin insulating
525 layer would need to be made of a material that is 36 times less conductive than PEEK (and hence 2324
526 times less conductive than stainless steel) to obtain the same degree of radial temperature gradient
527 reduction one could obtain if it would be possible to use an integral PEEK column.

528
529 Finally, comparing the black curve data (1mm column) with the other data (2.1 mm columns) in Fig. 12
530 shows that the most simple and straightforward approach to abating the viscous heating band broadening
531 still consists of resorting to thinner columns. Obviously, the field is not ready for this yet, given the ongoing
532 debate about the feasibility to pack high quality 1mm columns ^[28,29], as well as the requirement to move
533 to a micro-flow instrument in order to use these columns without extra-column band broadening losses.

534

535 **Conclusions**

536 Whereas the production of spherical core-shell particles with a high conductivity core can only be
537 expected to produce a relatively moderate increase in the bed conductivity, the simulation with the core-
538 shell monolithic structure conducted in the present study shows that the only way to use high conductivity
539 solid cores to effectively abate the viscous heating problem in UHPLC at pressures above 2000 bar is by
540 using structures where this solid core forms one uninterrupted path throughout the entire bed, as is the
541 case in the presently considered idealized tetrahedral skeleton model structure. Whether or not it will
542 ever be possible to produce such structures having the required small domain size (order μm) is however
543 an open question. Simulations conducted at the column level showed that the bed conductivities of such
544 structures can be high enough to alleviate the radial temperature gradients in the bed and offer a solution
545 to the viscous heating problem, while the bed conductivities attained by using spherical core-shell
546 particles with a high conductivity core were too low. An alternative approach, being the use of low
547 conductivity wall materials, was investigated. It was shown that this approach also led to reduced radial
548 temperature gradients. Besides practical considerations (e.g. mechanical strength), another downside of
549 this approach was that the lower ΔT_{rad} was inevitably accompanied by an increase in ΔT_{ax} . This is
550 disadvantageous, because it induces an extra loss in retention and unpredictable selectivity amongst
551 other effects ^[30,31]. As predicted by theory, the most simple and straightforward solution to abate viscous

552 heating obviously is a reduction of the column ID, as shown by the 1mm ID simulations conducted at the
553 end of this study. This solution however still requires the development of column packing procedures that
554 work as well as for 2.1mm ID columns^[28] and a new generation of instruments with further reduced extra-
555 column volumes and dispersion sources as well as with an increased pressure rating.

556

557 **Acknowledgements**

558 S. Deridder gratefully acknowledges a research grant from the Research Foundation – Flanders (FWO-
559 Vlaanderen).

560 G. Desmet and W. Smits gratefully acknowledge the support of ERC Advanced Grant nr. 695067.

561

562 **References**

563 ^[1] H.-J. Lin, S. Horvath, Viscous dissipation in packed beds, *Chem. Eng. Sci.* 36 (1981) 47-55.

564 DOI: 10.1016/0009-2509(81)80047-4

565 ^[2] H. Poppe, J.C. Kraak, J.F.K. Huber, J.H.M. van den Berg, Temperature-gradients in HPLC Columns
566 due to Viscous Heat Dissipation, *Chromatographia* 14 (1981) 515-523.

567 DOI: 10.1007/bf02265631

568 ^[3] H. Poppe, J.C. Kraak, Influence of thermal conditions on the efficiency of high-performance liquid
569 chromatographic columns, *J. Chromatogr.* 282 (1983) 399-412.

570 DOI: 10.1016/S0021-9673(00)91617-0

571 ^[4] U. Neue, M. Kele, Performance of idealized column structures under high pressure, *J. Chromatogr.*
572 *A* 1149 (2007) 236-244.

573 DOI: 10.1016/j.chroma.2007.03.042

574 ^[5] F. Gritti, M. Martin, G. Guiochon, Influence of viscous friction heating on the efficiency of columns
575 operated under very high pressures, *Anal. Chem.* 81 (2009) 3365-3384.

576 DOI: 10.1021/ac802632x

577 ^[6] K. Kaczmarski, F. Gritti, J. Kostka, G. Guiochon, Modeling of thermal processes in high pressure liquid
578 chromatography: II. Thermal heterogeneity at very high pressures, *J. Chromatogr. A* 1216 (2009)
579 6575-6586.

580 DOI: 10.1016/j.chroma.2009.07.049

581 ^[7] K. Broeckhoven, G. Desmet, Considerations for the use of ultra-high pressures in liquid
582 chromatography for 2.1 mm inner diameter columns, *J. Chromatogr. A* 1523 (2017) 183-192.

583 DOI: 10.1016/j.chroma.2017.07.040

584 ^[8] L. Tolley, J.W. Jorgenson, M.A. Moseley, Very High Pressure Gradient LC/MS/MS, *Anal. Chem.* 73
585 (2001) 2985-2991.

586 DOI: 10.1021/ac0010835

587 ^[9] R.E. Majors, Current Trends in HPLC Column Usage, LC GC Eur. 25 (2012) 31-39.
588 DOI:/

589 ^[10] S. Deridder, W. Smits, H. Benkahla, K. Broeckhoven, G. Desmet, Numerical and analytical
590 investigation of the possibilities to enhance the thermal conductivity of core-shell particle packed
591 beds, J. Chromatogr. A 1575 (2019) 26-33.
592 DOI: 10.1016/j.chroma.2018.08.056

593 ^[11] N. Vervoort, P. Gzil, G.V. Baron, G. Desmet, A correlation for the pressure drop in monolithic silica
594 columns, Anal. Chem. 75 (2003) 843-850.
595 DOI: 10.1021/ac0262199

596 ^[12] Y.A. Çengel, Heat transfer: a practical approach, 2nd ed., McGraw-Hill, New York, 2003.

597 ^[13] S. W. Churchill, H. H. S. Chu, Correlating equations for laminar and turbulent free convection from
598 a horizontal cylinder, Int. J. Heat Mass Tran. 18 (1975) 1049–1053.
599 DOI: 10.1016/0017-9310(75)90222-7

600 ^[14] J. Billen, K. Broeckhoven, A. Liekens, K. Choiket, G. Rozing, G. Desmet, Influence of pressure and
601 temperature on the physico-chemical properties of mobile phase mixtures commonly used in high-
602 performance liquid chromatography, J. Chromatogr. A 1210 (2008) 30-44.
603 DOI: 10.1016/j.chroma.2008.09.056

604 ^[15] R.J. Hulse, M.W. Anderson, M.D. Bybee, D.D. Gonda, C.A. Miller, J.L. Oscarson, R.L. Rowley, W.V.
605 Wilding, Liquid Thermal Conductivities of Acetonitrile, Diethyl Sulfide, Hexamethyleneimine,
606 Tetrahydrothiophene and Tetramethylethylenediamine, J. Chem. Eng. Data 49 (2004) 1433-1435.
607 DOI: 10.1021/je0498661

608 ^[16] A.M. Kolker, L.P. Safonova, Molar heat capacities of the (water + acetonitrile) mixtures at T=
609 (283.15, 298.15, 313.15, and 328.15) K, J. Chem. Thermodyn. 42 (2010) 1209-1212.
610 DOI: 10.1016/j.jct.2010.04.019

611 ^[17] D.T. Jamieson, J. B. Irving, J. S. Tudhope, Liquid Thermal Conductivity. A Data Survey to 1973, H. M.
612 Stationary Office, Edinburgh, 1975.

613 ^[18] M. Martin, G. Guiochon, Effects of high pressure in liquid chromatography, J. Chromatogr. A 1090
614 (2005) 16-38.
615 DOI: 10.1016/j.chroma.2005.06.005

616 ^[19] F. Gritti, G. Guiochon, Complete temperature profiles in ultra-high-pressure liquid chromatography
617 columns, Anal. Chem. 80 (2008) 5009-5020.
618 DOI: 10.1021/ac800280c

619 ^[20] D. Cabooter, K. Broeckhoven, R. Sterken, A. Vanmessen, I. Vandendael, K. Nakanishi, S. Deridder, G.
620 Desmet, Detailed characterization of the kinetic performance of first and second generation silica
621 monolithic columns for reversed-phase chromatography separations, *J. Chromatogr. A* 1325 (2014)
622 72-82.
623 DOI: 10.1016/j.chroma.2013.11.047

624 ^[21] D. Åsberg, M. Chutkowski, M. Lesko, J. Samuelsson, K. Kaczmarek, T. Fornstedt, A practical
625 approach for predicting retention time shifts due to pressure and temperature gradients in ultra-
626 high-pressure liquid chromatography, *J. Chromatography A* 1479 (2017) 107-120.
627 DOI: 10.1016/j.chroma.2016.11.050

628 ^[22] K. Kaczmarek, J. Kostka, W. Zapała, G. Guiochon, Modeling of thermal processes in high pressure
629 liquid chromatography I. Low pressure onset of thermal heterogeneity, *J. Chromatogr. A* 1216
630 (2009) 6560-6574.
631 DOI: 10.1016/j.chroma.2009.07.020

632 ^[23] K. Kaczmarek, F. Gritti, G. Guiochon, Prediction of the influence of the heat generated by viscous
633 friction on the efficiency of chromatography columns, *J. Chromatogr. A* 1177 (2008) 92-104.
634 DOI: 10.1016/j.chroma.2007.11.009

635 ^[24] F. Gritti, G. Guiochon, The current revolution in column technology: How it began, where is it
636 going?, *J. Chromatogr. A* 1228 (2012) 2-19.
637 DOI: 10.1016/j.chroma.2011.07.014

638 ^[25] J. P. Grinias, D. S. Keil, J. W. Jorgenson, Observation of enhanced heat dissipation in columns packed
639 with superficially porous particles, *J. Chromatogr. A* 1371 (2014) 261-264.
640 DOI: 10.1016/j.chroma.2014.10.075

641 ^[26] J. García Ten, M.J. Orts, A. Saburit, G. Silva, Thermal conductivity of traditional ceramics. Part I:
642 Influence of bulk density and firing temperature, *Ceram. Int.* 36 (2010) 1951-1959.
643 DOI: 10.1016/j.ceramint.2010.05.012

644 ^[27] F. Gritti, Vacuum-Jacketed Columns: Maximum Efficiency, Easy Deployment Without Oven, and
645 Improved LC-MS Performance, *LC GC Eur.* 32 (2019) 8-13.
646 DOI:/

647 ^[28] F. Gritti, M.F. Wahab, Understanding the Science Behind Packing High-Efficiency Columns and
648 Capillaries: Facts, Fundamentals, Challenges, and Future Directions, *LC GC Eur.* 31 (2018) 90-101.
649 DOI:/

650 ^[29] J.A. Anspach, T.D. Maloney, L.A. Colón, Ultrahigh-pressure liquid chromatography using a 1-mm id
651 column packed with 1.5- μm porous particles, *J. Sep. Sci.* 30 (2007) 1207-1213.
652 DOI: 10.1002/jssc.200600535

- 653 ^[30] K. Horváth, S. Horváth, D. Lukács, Effect of axial temperature gradient on chromatographic
654 efficiency under adiabatic conditions, J. Chromatogr. A 1483 (2017) 80-85.
655 DOI: 10.1016/j.chroma.2016.12.063
- 656 ^[31] S. Fekete, D. Guillarme, Estimation of pressure-, temperature- and frictional heating-related effects
657 on proteins' retention under ultra-high-pressure liquid chromatographic conditions, J. Chromatogr.
658 A 1393 (2015) 73-80.
659 DOI: 10.1016/j.chroma.2015.03.023

Figure captions

660

661

662 **Figure 1.** Unit cells of different ordered sphere packings **(a-c)**, unit cell of the random sphere packing
663 considered in Section 3.2 **(d)** and unit cells of the tetrahedral skeleton model geometries considered in
664 Sections 3.1 and 3.2 **(e,f)**. All sphere and branch radii are 1.00 μm . The corresponding unit cell lengths (in
665 direction of the temperature gradient) and porosities (ε) are indicated in the figure. Core material: black,
666 porous zone: grey, mobile zone: transparent blue. In case of fully porous structures, the core material was
667 assigned the same properties as the porous zone (shell).

668

669 **Figure 2.** Effective conductivity for porous layers made up of TSM structures (\bullet) or inversed TSM
670 structures (\diamond), for a range of porosities and different mobile phase conductivities. Red cubes (\blacksquare) are FCC,
671 BCC, SC packings at closest packing porosity and a random packing ($\varepsilon_e=0.39$) for $k_m=0.21$ W/(m.K). Blue
672 cubes (\blacksquare) are same packings with $k_m=0.58$ W/(m.K).

673

674 **Figure 3.** Flow-dependency of relative effective conductivity ($k_{\text{eff}}/k_{\text{eff},u=0}$) of a random packing of fully
675 porous spherical particles ($\varepsilon_e=0.39$). Conditions: $k_m=0.58$ W/(m.K), $k_{p_z}=0.80$ W/(m.K). The dashed lines
676 indicates $k_{\text{eff}}/k_{\text{eff},u=0} = 1$ (i.e. no influence of convection).

677

678 **Figure 4.** Effective conductivity (k_{eff}) of packed beds of spheres (perfectly ordered fcc-packing with $\varepsilon_e=0.24$
679 **(a)** and random packing with $\varepsilon_e=0.39$ **(b)**). Both fully porous (black data) and core-shell (gray data) particles
680 are considered. Conditions: $k_m=0.58$ W/(m.K), $k_{\text{SiO}_2}=1.4$ W/(m.K), $k_{\text{ZrO}_2}=2$ W/(m.K), $k_{\text{Al}_2\text{O}_3}=40$ W/(m.K),
681 $k_{\text{Cu}}=400$ W/(m.K).

682

683 **Figure 5.** Effective conductivity (k_{eff}) of monolithic beds (TSM-packing with $\varepsilon_e=0.39$ **(a)**, TSM-packing with
684 $\varepsilon_e=0.60$ **(b)**). Both fully porous and core-shell monoliths are considered (as indicated in figure).
685 Conditions: $k_m=0.58$ W/(m.K), $k_{\text{SiO}_2}=1.4$ W/(m.K), $k_{\text{Al}_2\text{O}_3}=40$ W/(m.K), $k_{\text{Cu}}=400$ W/(m.K). The curve
686 representing the data of zirconia core particles, coincides with the curve of the silica core case and was
687 left out.

688

689 **Figure 6.** Effective conductivity (k_{eff}) in a random packing of core-shell particles with varying core
690 conductivity k_{core} for pure water ($k_m=0.61$ W/(m.K), \blacksquare), pure ACN ($k_m=0.19$ W/(m.K), \blacktriangle) and a
691 40/60(v%/v%) ACN/H₂O mixture ($k_m=0.41$ W/(m.K), \bullet).

692

693 **Figure 7.** Temperature distribution profiles in a 50x2.1mm metal tubing column operated at 2500bar for
694 **(a)** isothermal, **(b)** adiabatic and **(c)** natural convection boundary conditions imposed at the outer column

695 walls. Flow rate=0.5mL/min, mobile phase=pure water. Arrows in **(a)** define the difference between
696 $\Delta T_{ax,bed}$ and $\Delta T_{ax,col}$ used in the text. Arrows in **(c)** indicate the main column dimensions (drawing to scale).
697

698 **Figure 8.** Plots of $\Delta T_{rad,bed}$ as a function of **(a)** $\Delta T_{ax,col}$ and **(b)** $\Delta T_{ax,bed}$ in a 2.1x50mm metal tubing column
699 operated at different column pressures (500, 750, 1000, 1500, 2000 and 2500bar) with a 40/60(v%/v%)
700 ACN/H₂O mixture and $k_{eff}=0.59$ W/(m.K). Three different thermal boundary conditions as indicated in
701 figure: isothermal, adiabatic and natural convection (still-air). Flow rate=0.5mL/min for adiabatic case.
702 Bed permeabilities of the adiabatic data points were also used for the corresponding (same inlet pressure)
703 natural convection and isothermal points. Different column inlet pressures relating to the different data
704 points are indicated in figure and follow the same order for all curves.
705

706 **Figure 9.** Effect of liquid composition on plots of $\Delta T_{rad,bed}$ as a function of $\Delta T_{ax,bed}$ in a 2.1x50mm metal
707 tubing column operated at different column pressures (500, 750, 1000, 1500, 2000 and 2500bar) for **(a)**
708 natural convection (still-air) and **(b)** adiabatic boundary conditions. Considered liquids: pure water (■),
709 pure ACN (▲) and a 40/60(v%/v%) ACN/H₂O mixture (●). Flow rate=0.5mL/min for every pressure
710 considered for the ACN/H₂O mixture under adiabatic conditions. Flow rates for natural convection
711 conditions and for pure water and ACN changed in relation to their different viscosity (see text for details).
712 Values for k_{eff} : 0.77 W/(m.K) (■), 0.37 W/(m.K) (▲) and 0.59 W/(m.K) (●).
713

714 **Figure 10.** Evolution of $\Delta T_{rad,bed}$ as a function of the column pressure drop for natural convection (full
715 symbols) and adiabatic (open symbols) boundary conditions. Considered liquids: pure water (■), pure
716 ACN (▲) and a 40/60(v%/v%) ACN/H₂O mixture (●). (same data as in figure 9, different representation)
717

718 **Figure 11.** Effect of increases in k_{eff} on the $\Delta T_{rad,bed}$ versus $\Delta T_{ax,bed}$ -relationship under natural convection
719 (still air) conditions for **(a)** a 40/60 ACN/water-mixture and **(b)** pure ACN. Different column inlet pressures
720 are indicated in figure and follow the same order for all curves.
721

722 **Figure 12.** Effect of the conductivity of the column tube wall (k_{wall}) on the $\Delta T_{rad,bed}$ versus $\Delta T_{ax,bed}$ -
723 relationship under **(a)** natural convection (still air) and **(b)** adiabatic conditions for the 40/60 ACN/water-
724 mixture. Red data represent single-material columns with either a regular wall conductivity ($k_{wall}=k_{stainless}$
725 $_{steel}=16.27$ W/(m.K), solid line) or a lowered wall conductivity (dashed lines). Yellow data represent a
726 stainless steel column ($k_{wall}=16.27$ W/(m.K)) with extra internal coating (100 μ m thickness) with different
727 thermal conductivities (A: $k_{coat}=0.25$ W/(m.K), B: $k_{coat}=0.0625$ W/(m.K), C: $k_{coat}=0.0156$ W/(m.K), D:
728 $k_{coat}=0.0039$ W/(m.K)). Black data represent a regular column with 1mm ID. Different column inlet
729 pressures are indicated in figure and are the same for all curves, with one extra high pressure (3500bar)
730 in case of the 1mm ID column.

Fig. 1:

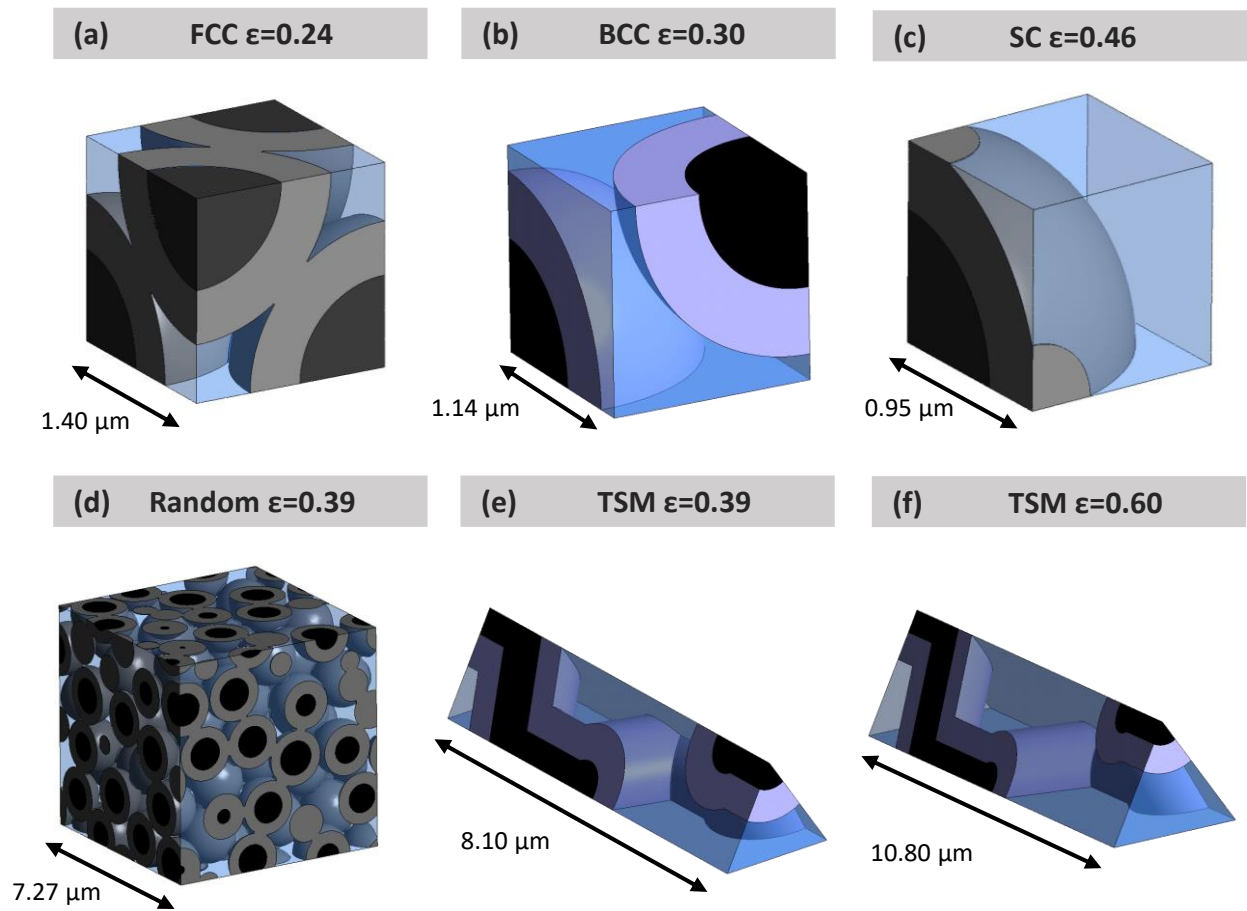


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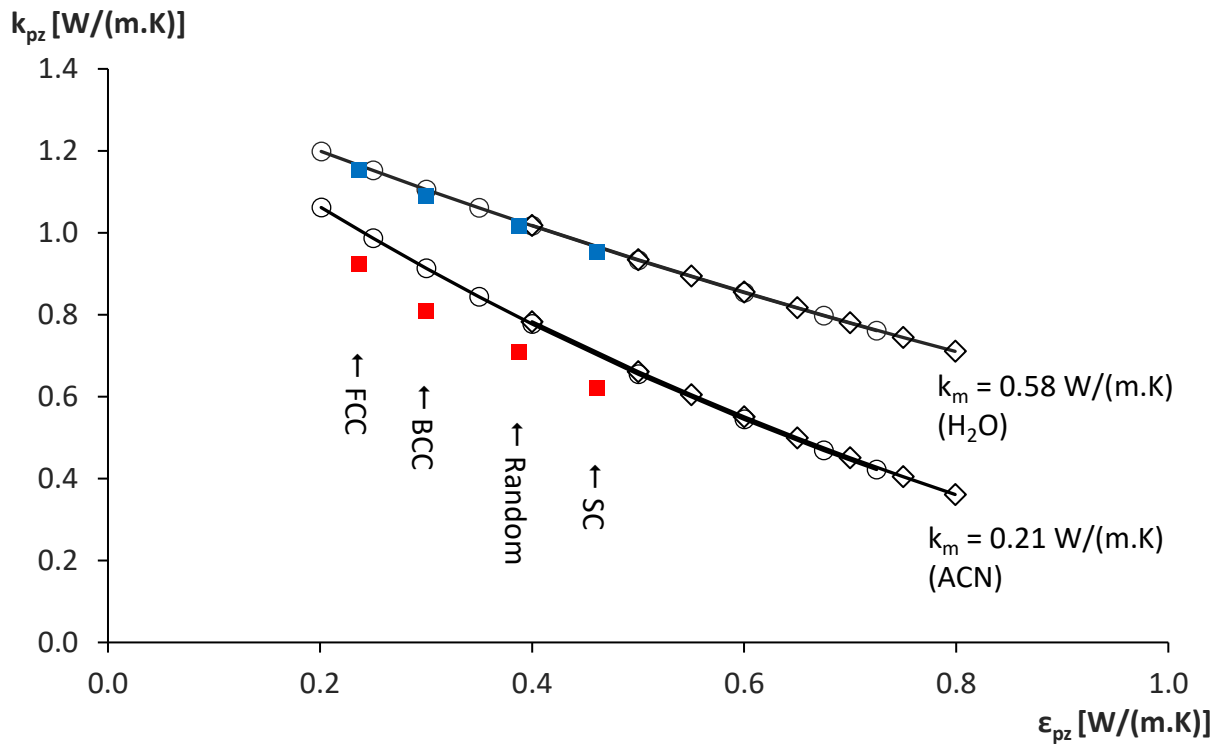


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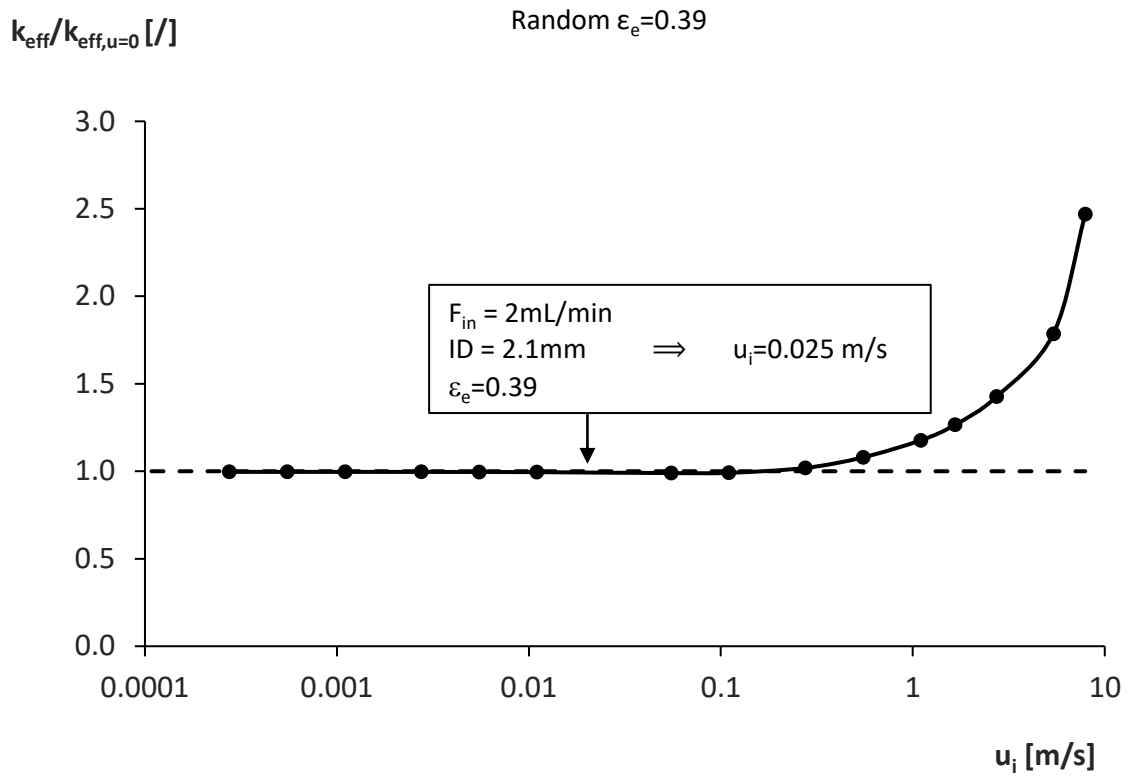


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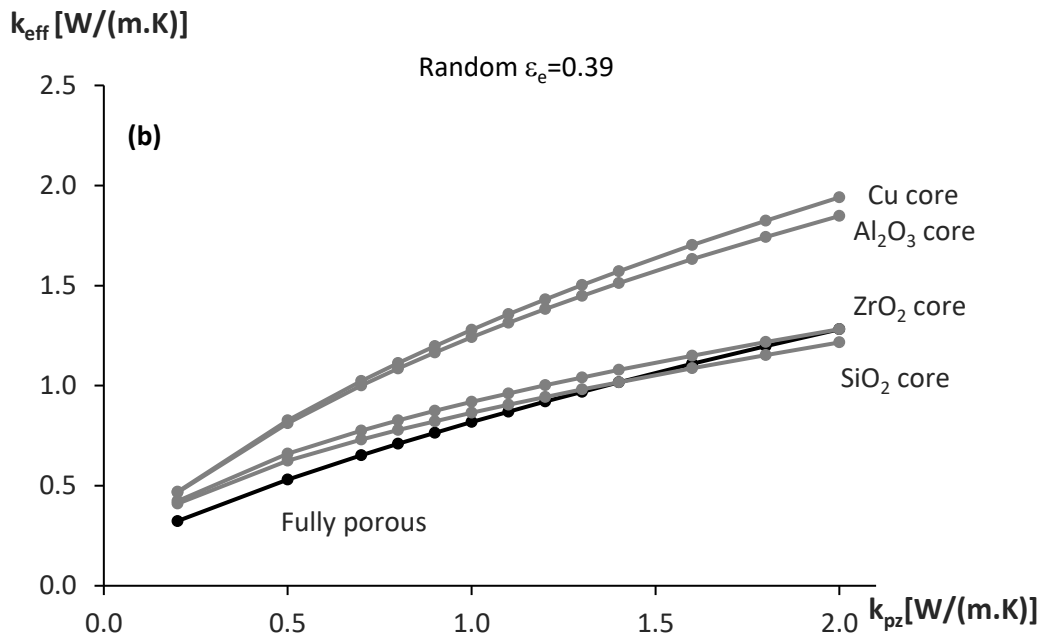
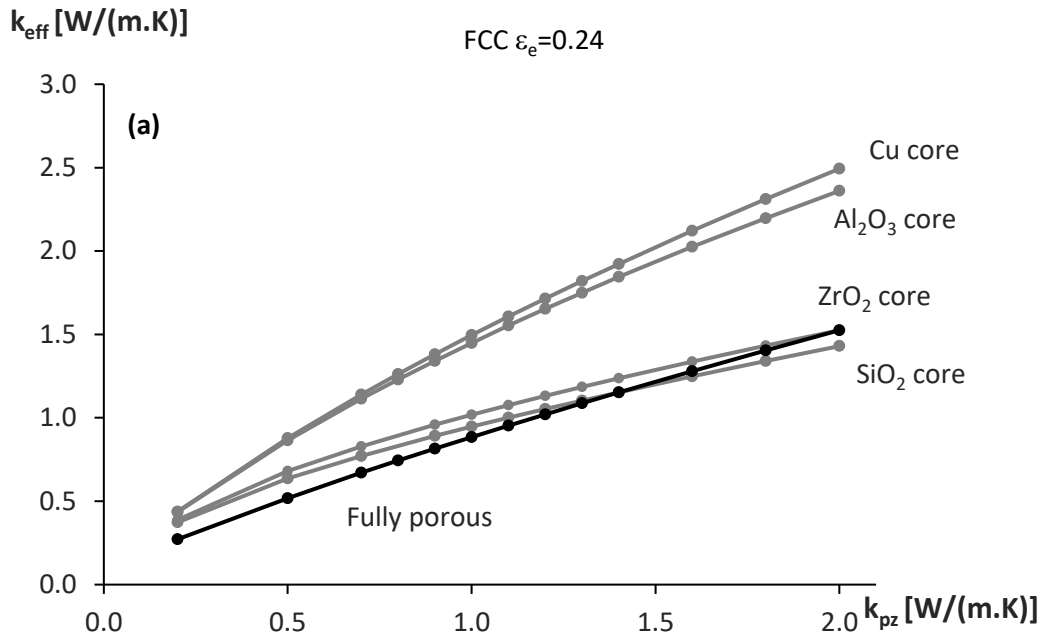


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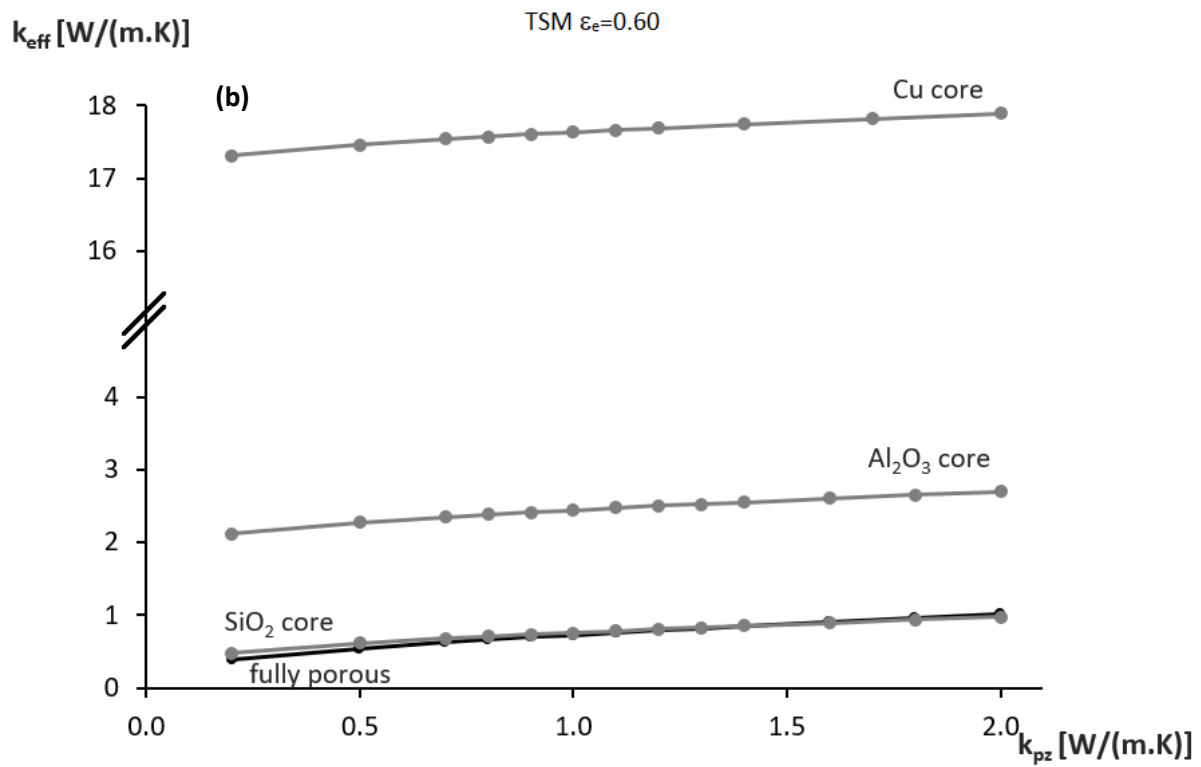
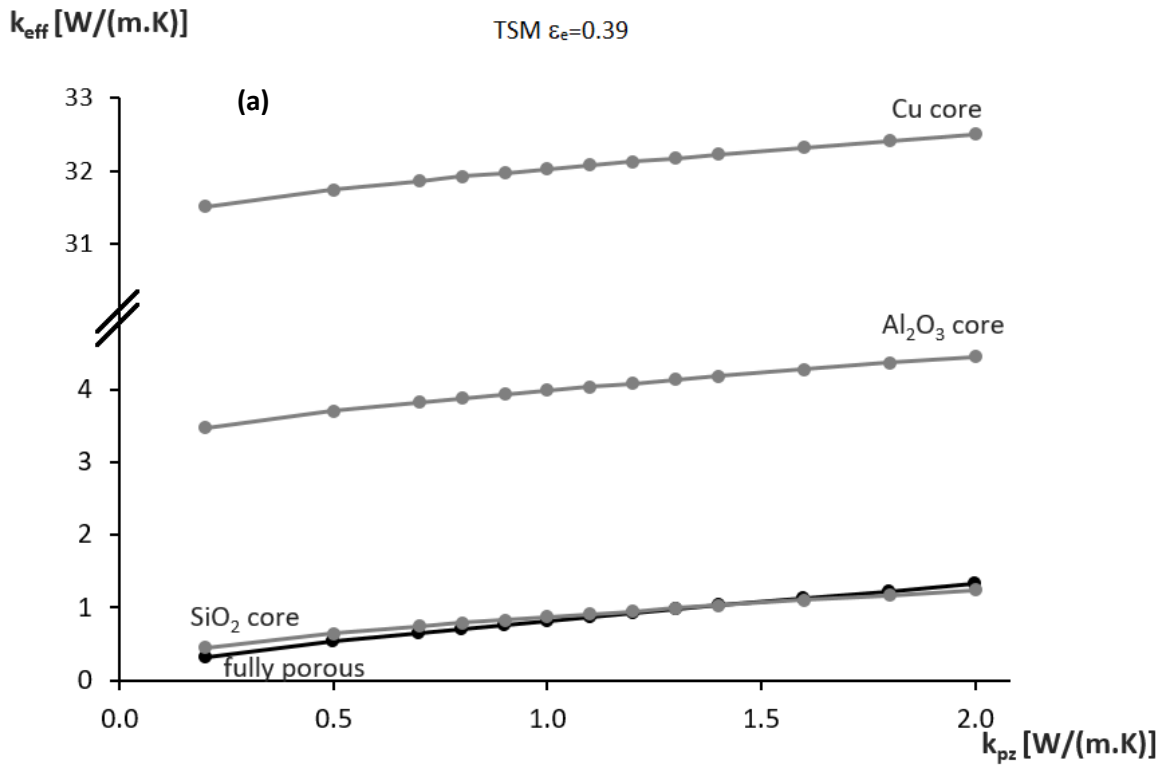


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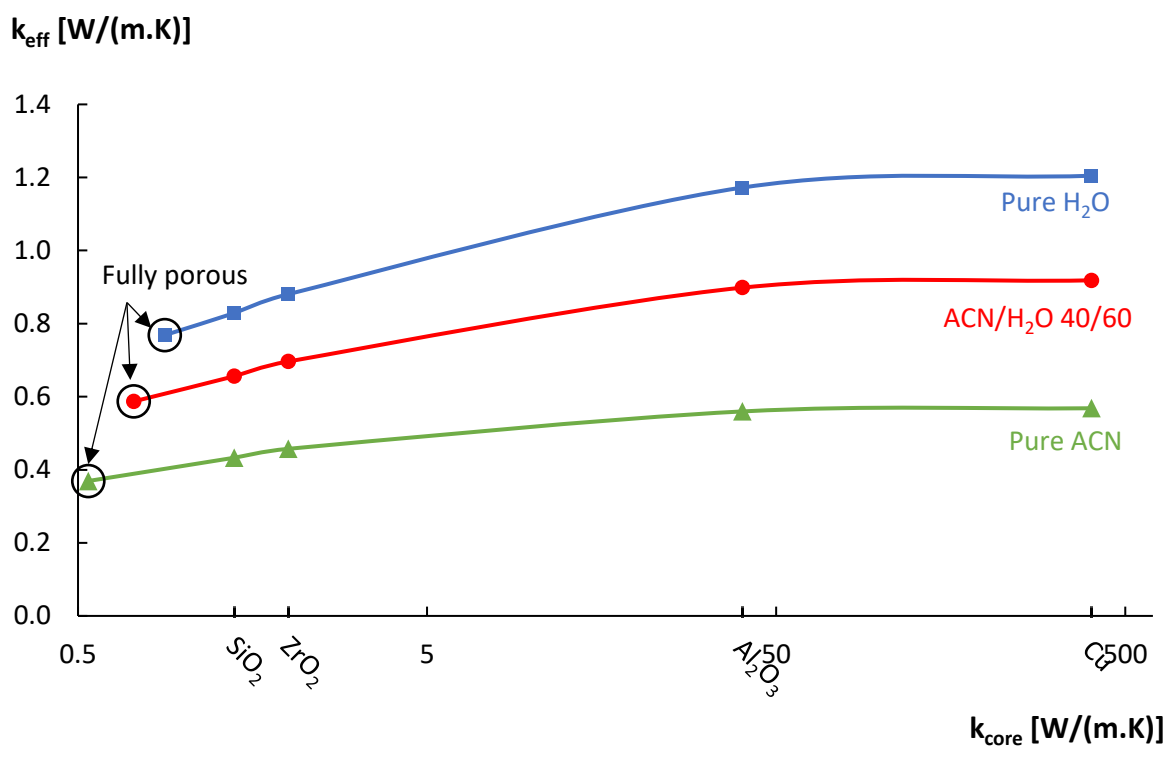


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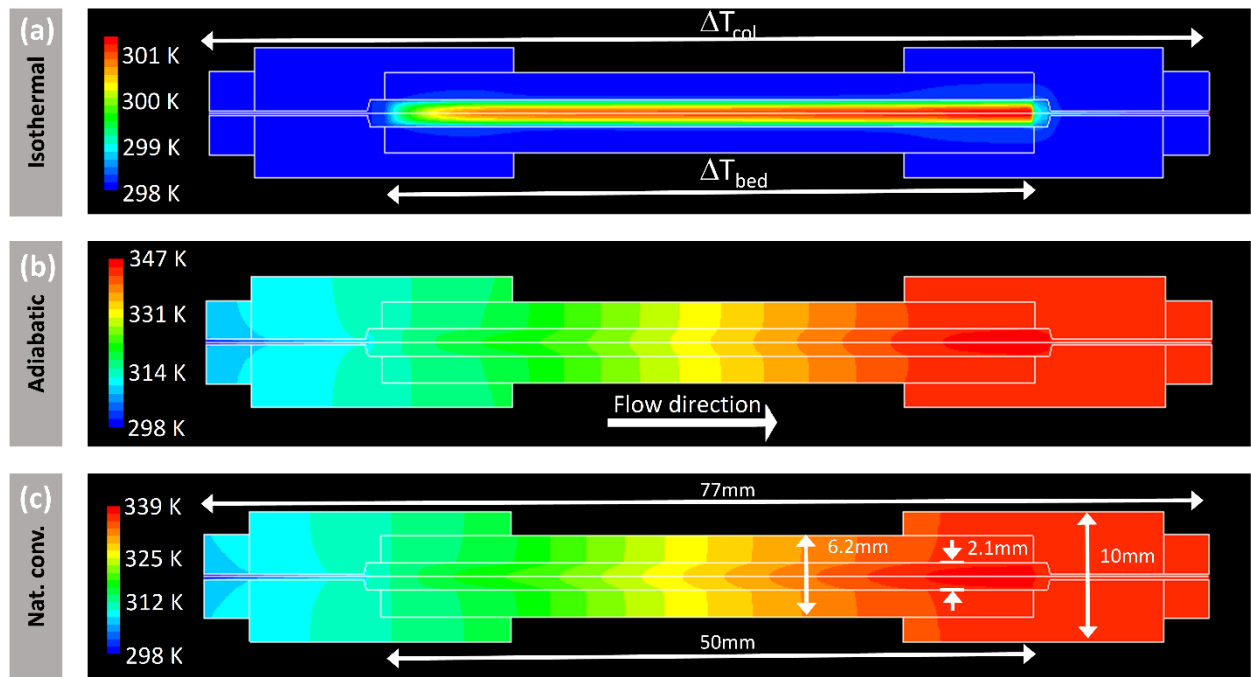


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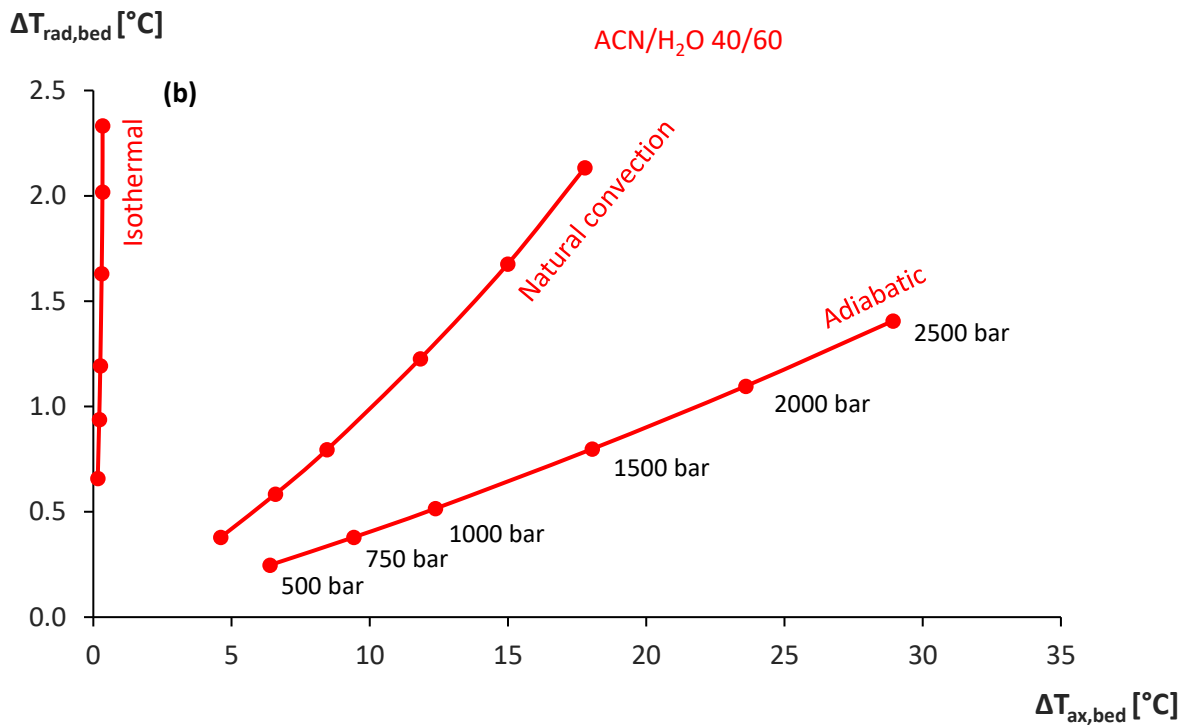
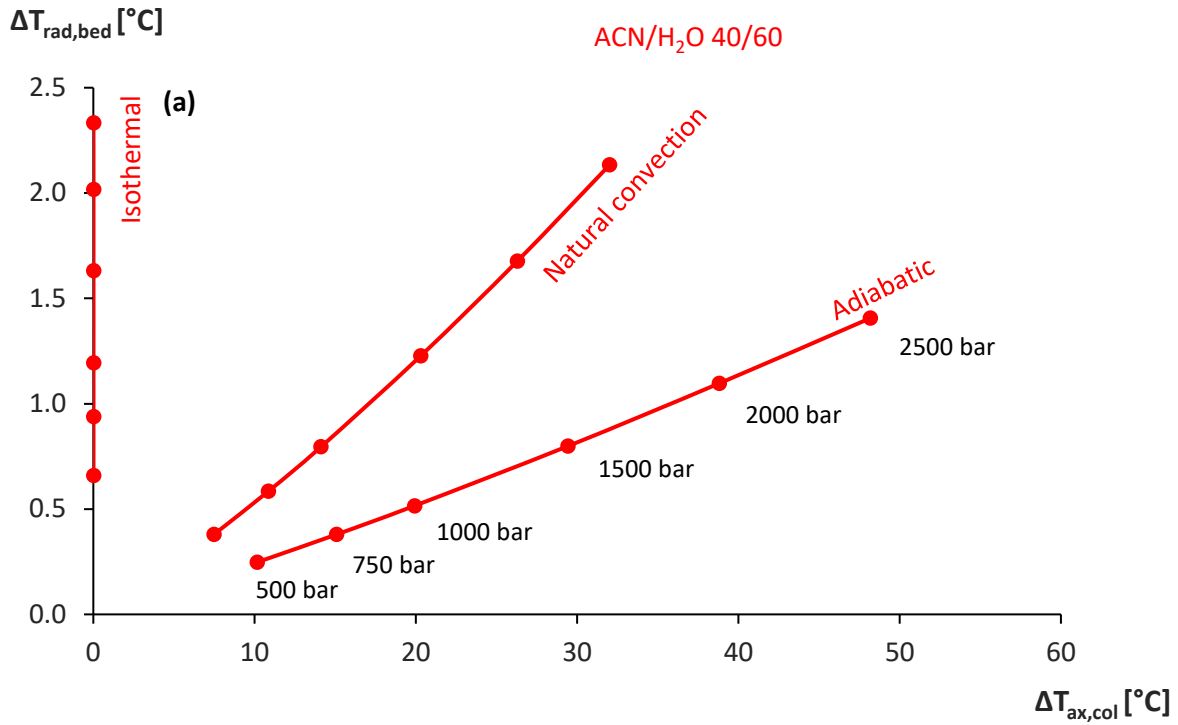


Fig. 9:

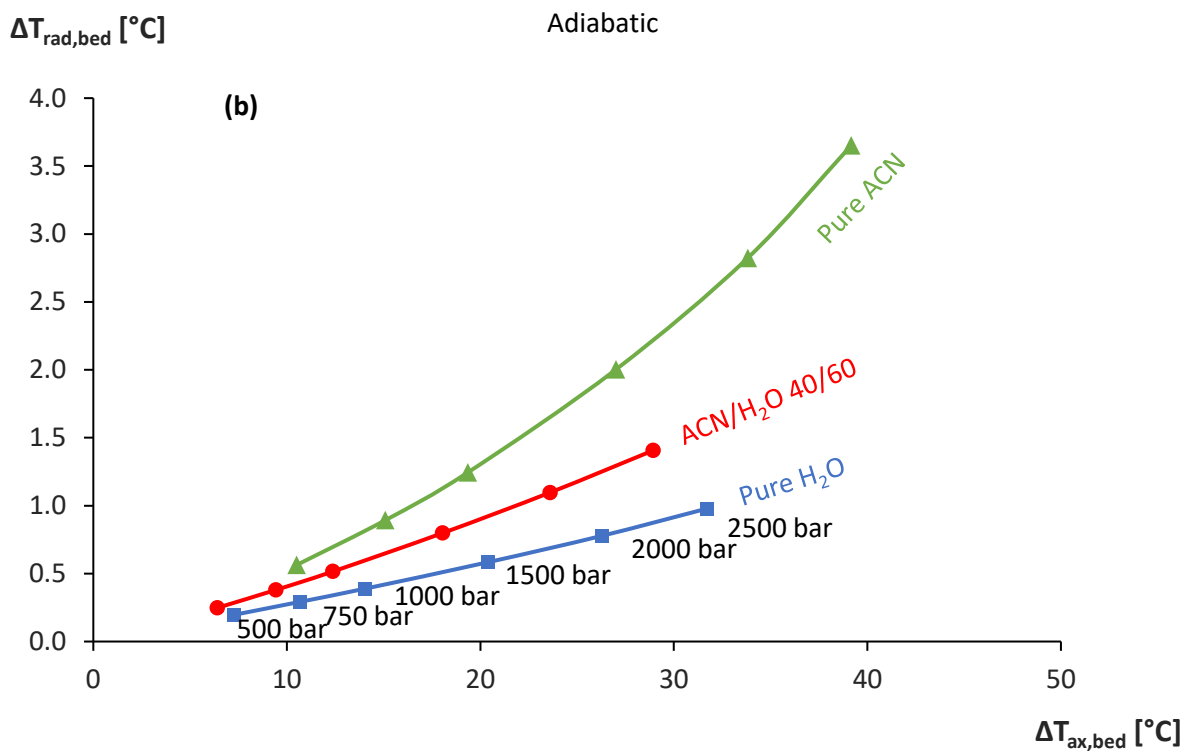
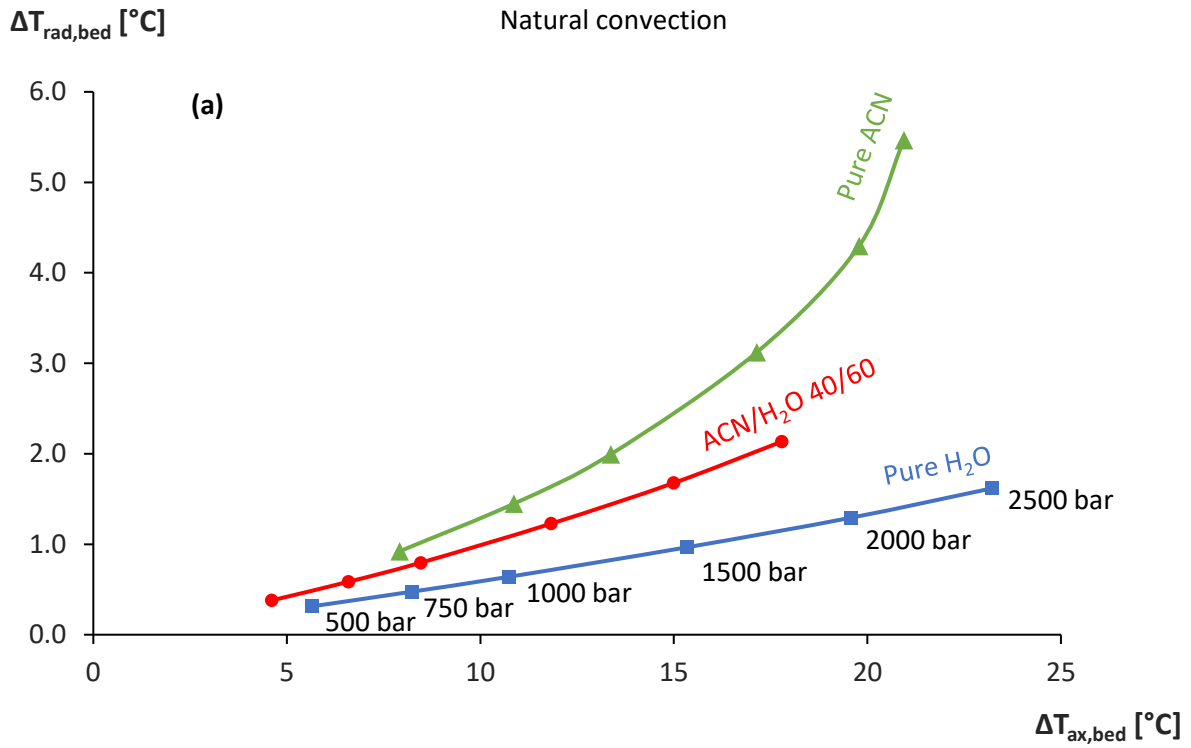


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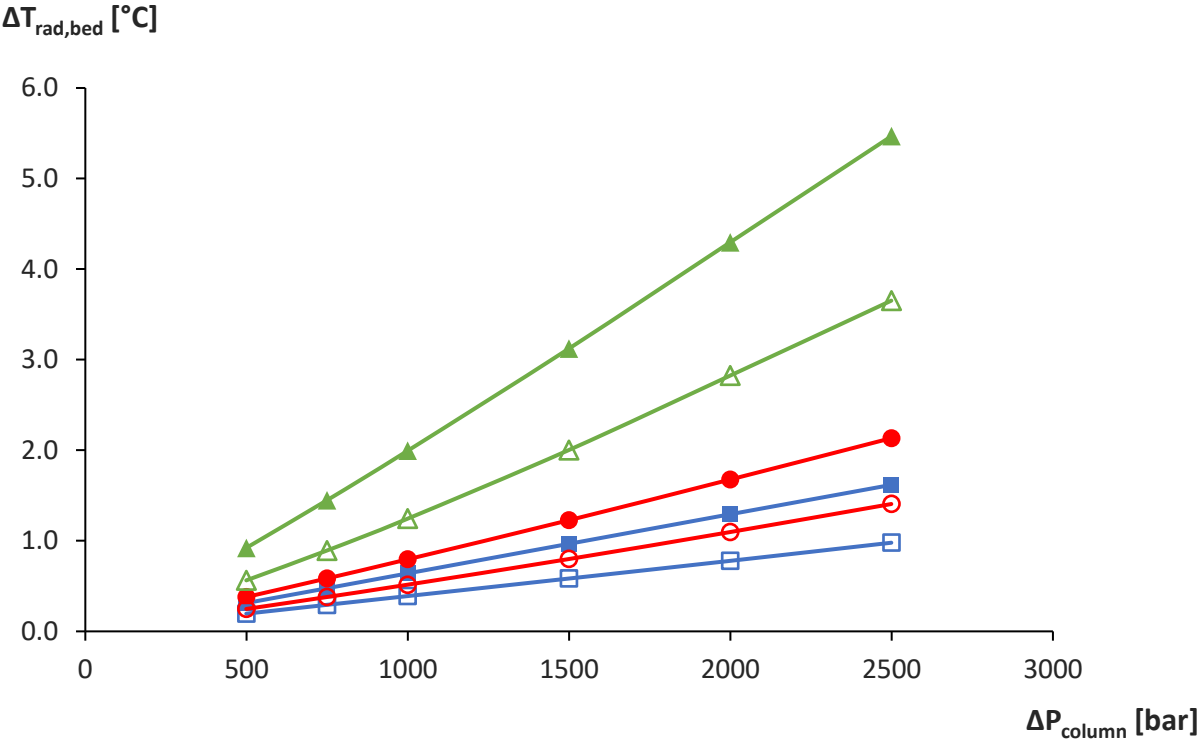


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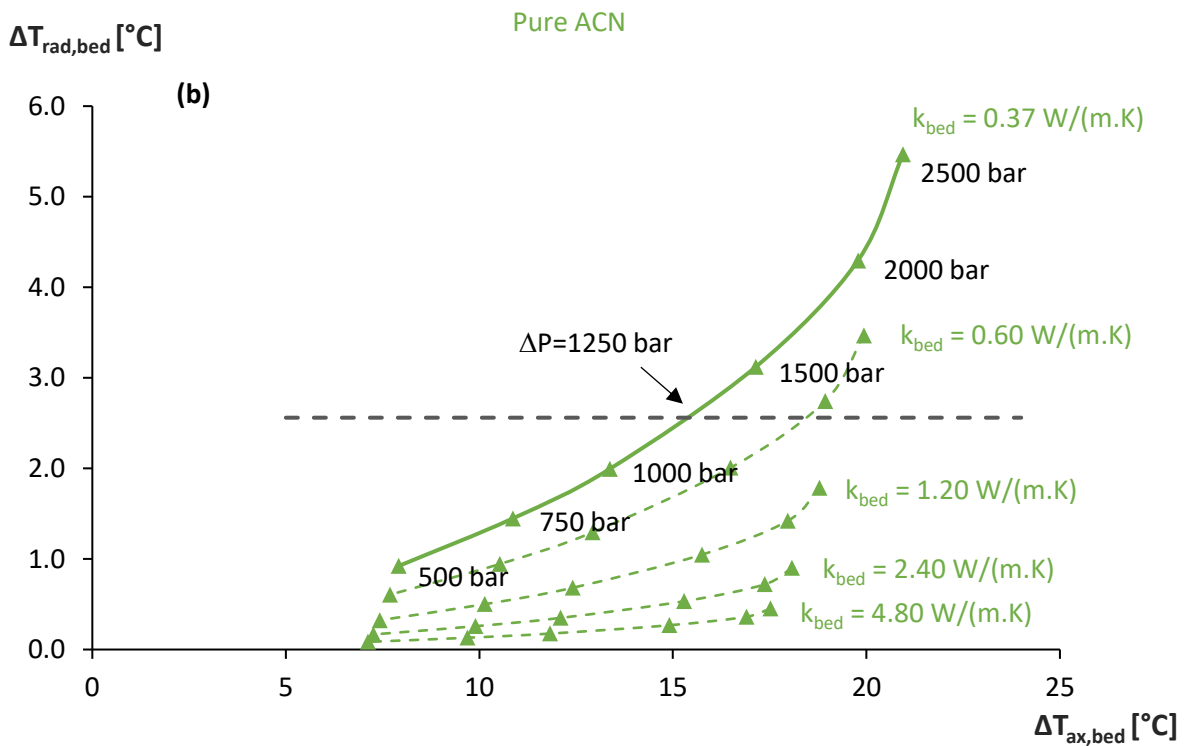
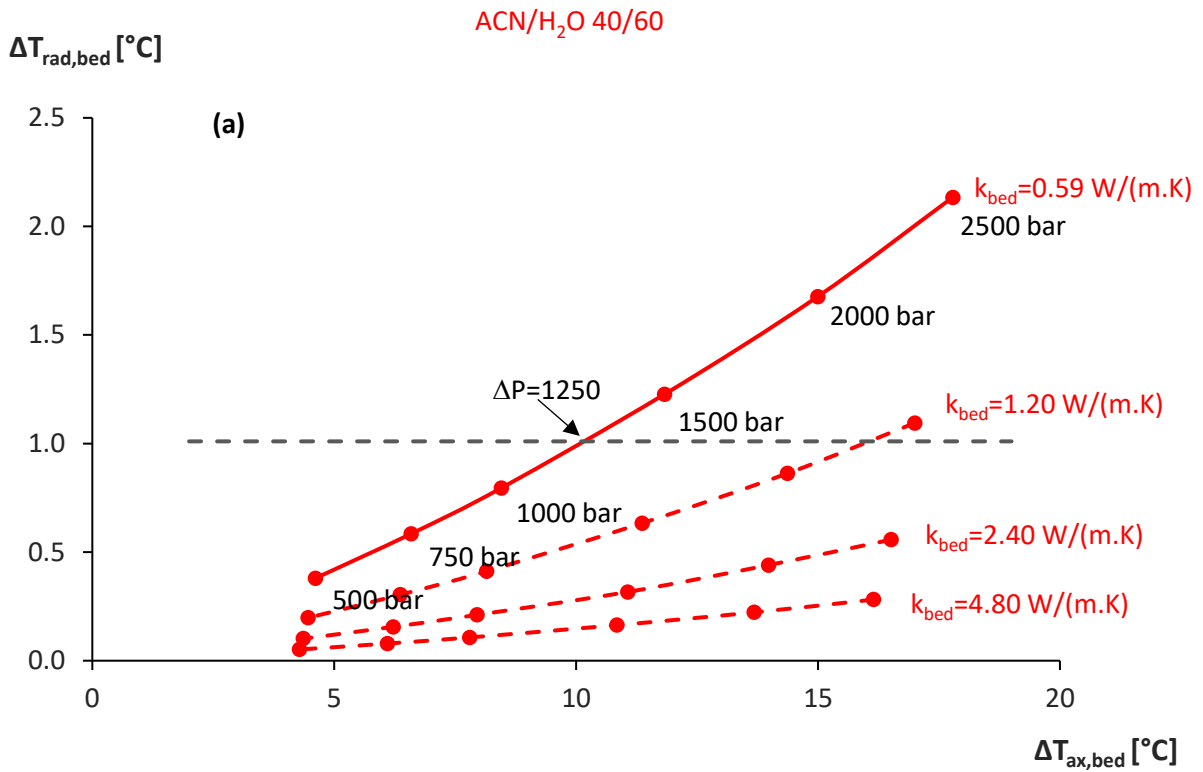
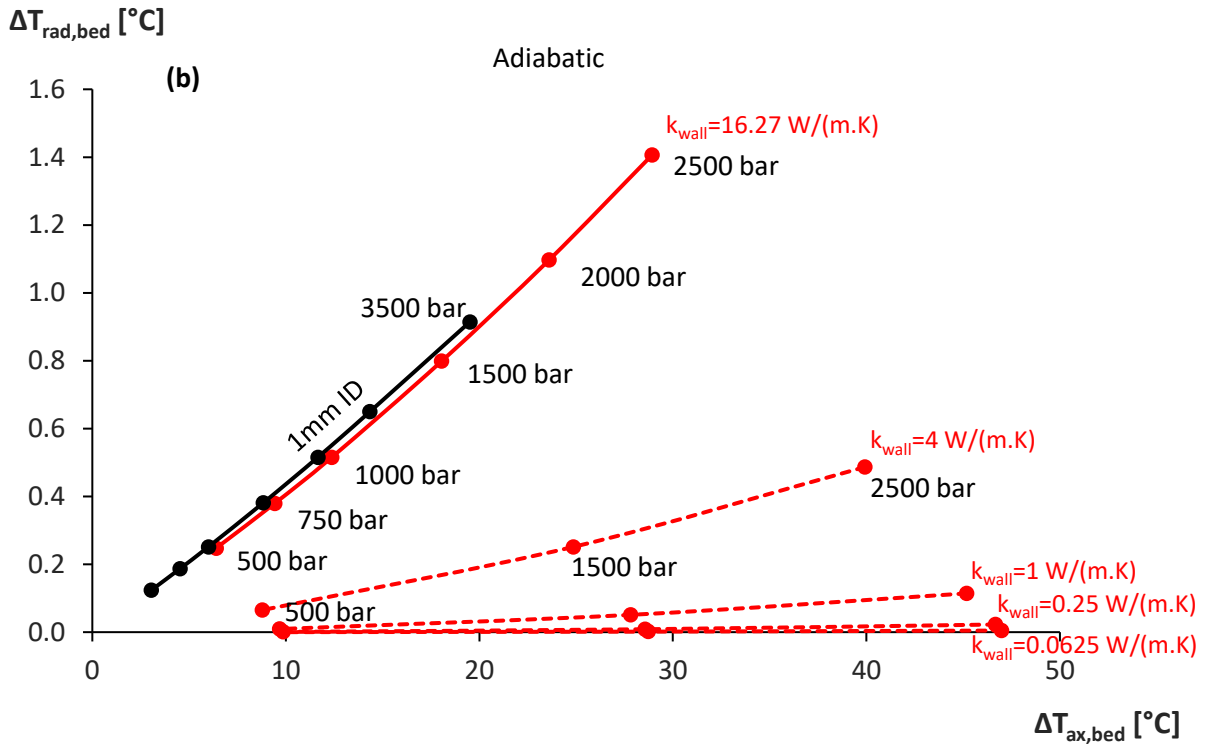
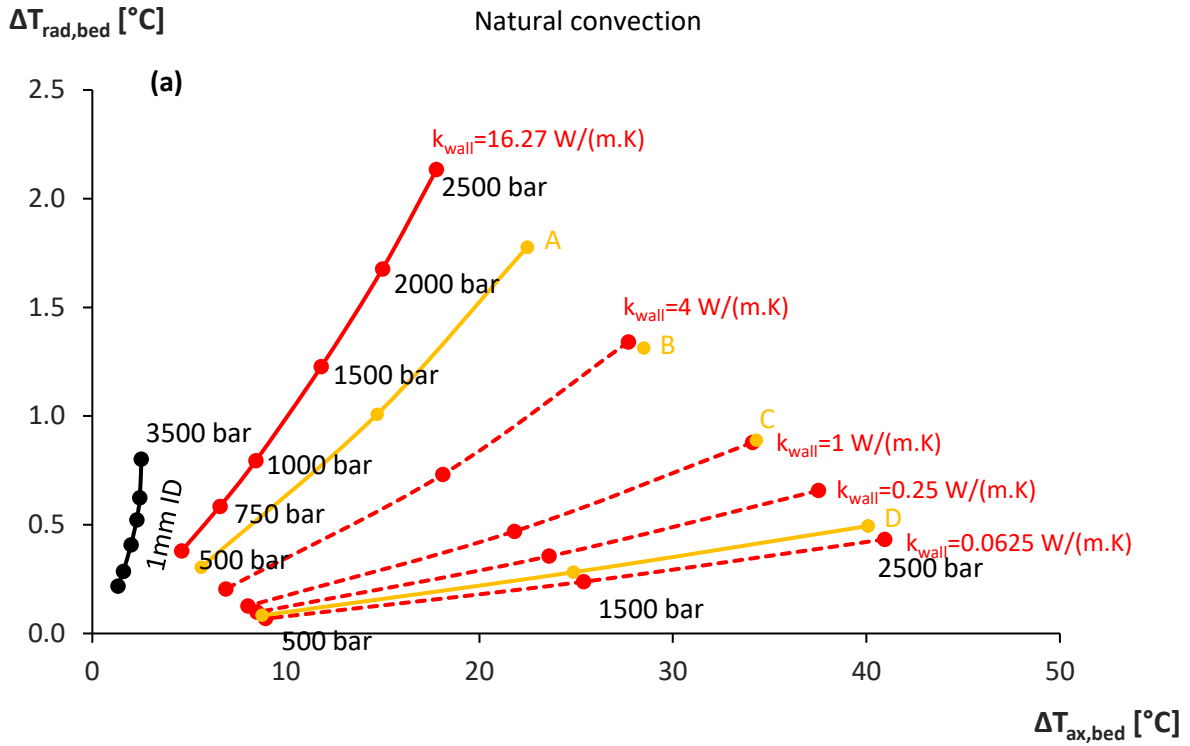


Fig. 12:



Supplementary information for:

A Multiscale Modelling Study on the Sense and Nonsense of Thermal Conductivity Enhancement of Liquid Chromatography Packings and Other Potential Solutions for Viscous Heating Effects

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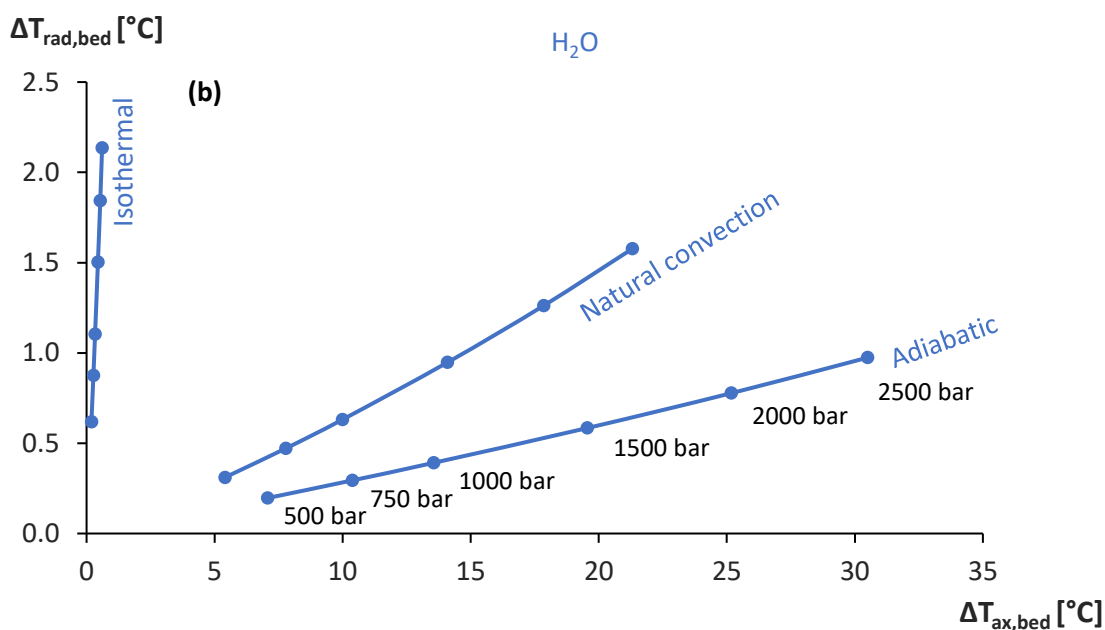
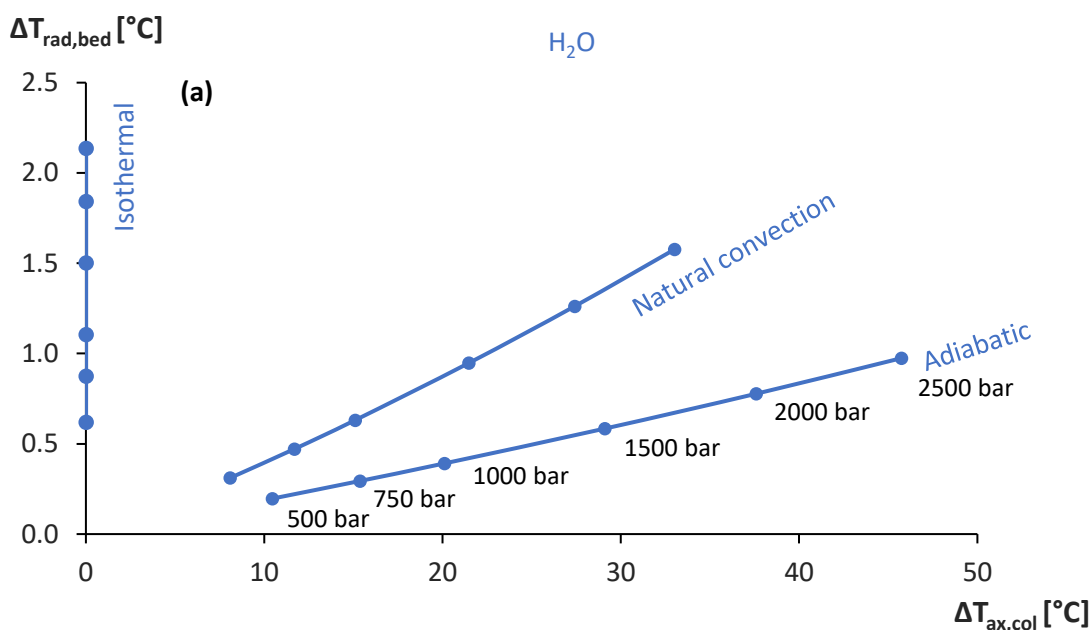


Figure S1. Plots of $\Delta T_{rad,bed}$ as a function of (a) $\Delta T_{ax,col}$ and (b) $\Delta T_{ax,bed}$ in a 2.1x50mm metal tubing column operated at different column pressures (500, 750, 1000, 1500, 2000 and 2500bar) with a pure H_2O mobile phase and $k_{eff}=0.77$ W/(m.K). Three different thermal boundary conditions as indicated in figure: isothermal, adiabatic and natural convection (still-air). Flow rate=0.5mL/min for adiabatic case. Bed permeabilities of the adiabatic data points were also used for the corresponding (same inlet pressure) natural convection and isothermal points. Different column inlet pressures are indicated in figure and are the same for all curves.

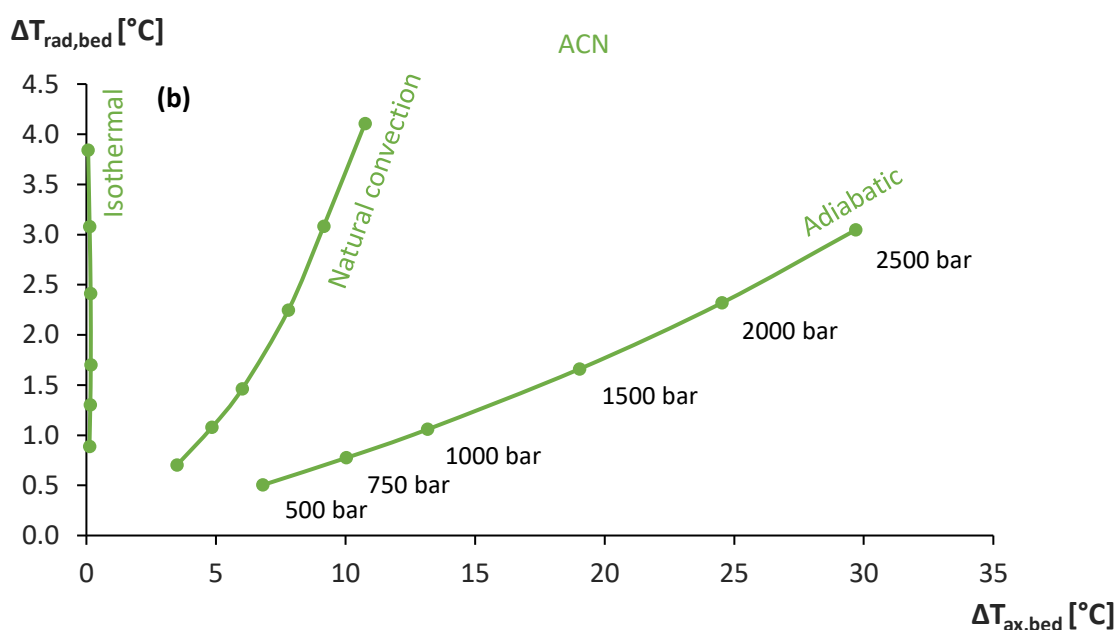
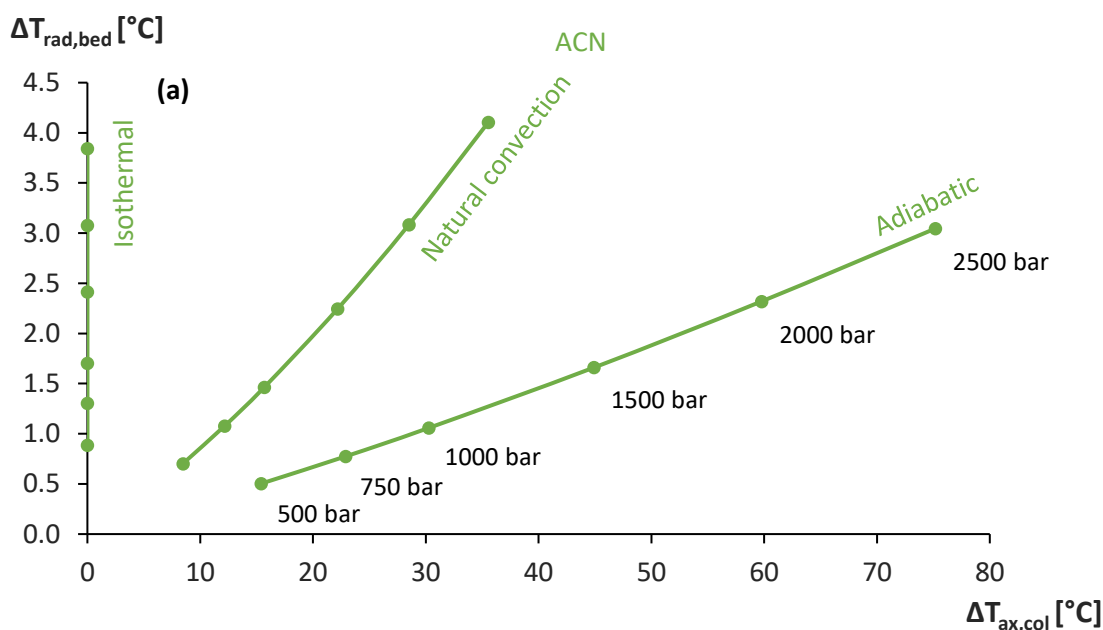


Figure S2. Plots of $\Delta T_{rad,bed}$ as a function of (a) $\Delta T_{ax,col}$ and (b) $\Delta T_{ax,bed}$ in a 2.1x50mm metal tubing column operated at different column pressures (500, 750, 1000, 1500, 2000 and 2500bar) with a pure ACN mobile phase and $k_{eff}=0.37$ W/(m.K). Three different thermal boundary conditions as indicated in figure: isothermal, adiabatic and natural convection (still-air). Flow rate=0.5mL/min for adiabatic case. Bed permeabilities of the adiabatic data points were also used for the corresponding (same inlet pressure) natural convection and isothermal points. Different column inlet pressures are indicated in figure and are the same for all curves.

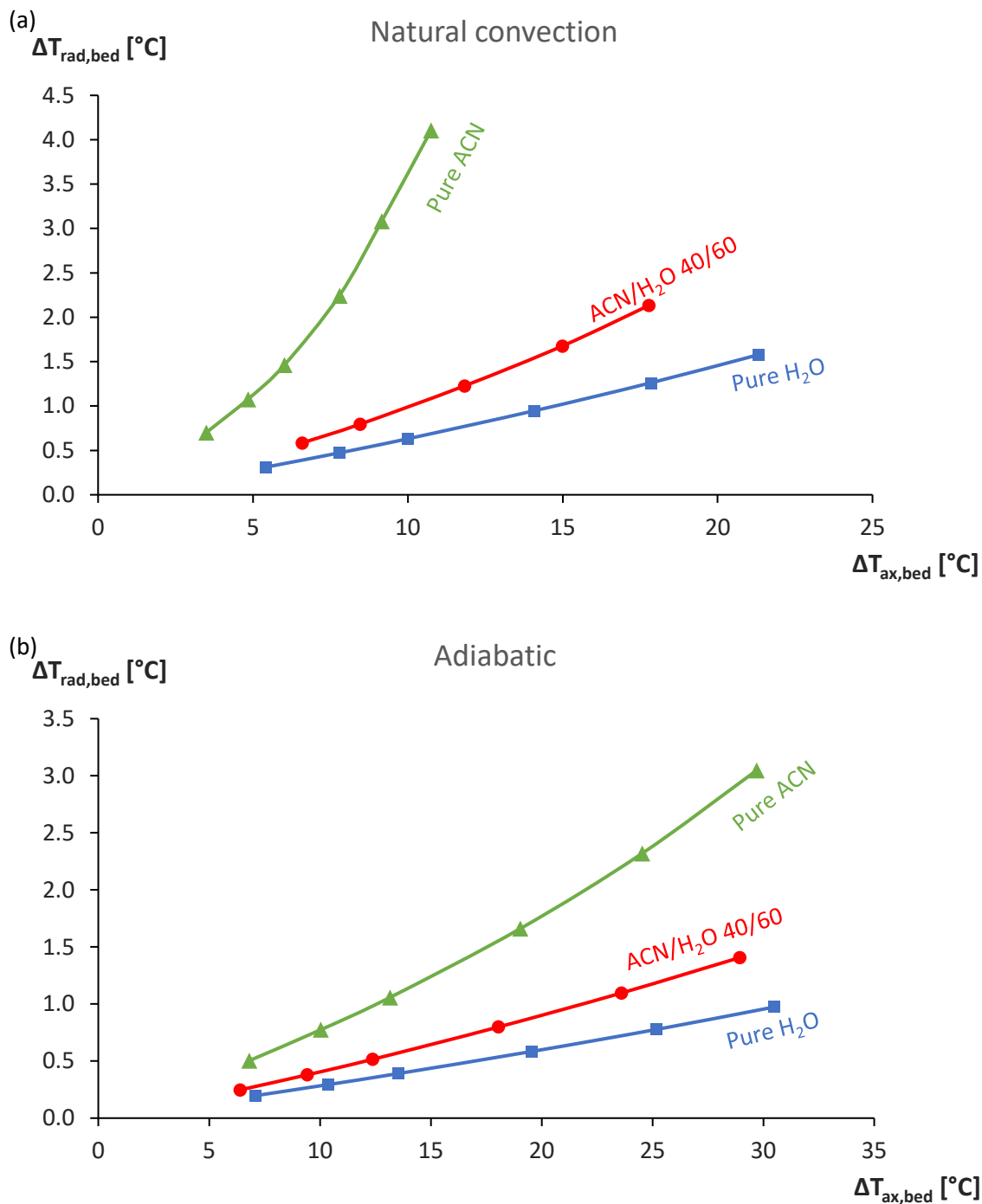


Figure S3. Effect of liquid composition on plots of $\Delta T_{\text{rad,bed}}$ as a function of $\Delta T_{\text{ax,bed}}$ in a 2.1x50mm metal tubing column operated at different column pressures (500, 750, 1000, 1500, 2000 and 2500bar) for (a) natural convection (still-air) and (b) adiabatic boundary conditions. Considered liquids: pure water (■), pure ACN (▲) and a 40/60(v%/v%) ACN/H₂O mixture (●). Flow rate=0.5mL/min under adiabatic conditions (for all inlet pressures considered and all three mobile phases). Flow rates for natural convection conditions changed in relation to their different viscosity. Values for k_{eff} : 0.77 W/(m.K) (■), 0.37 W/(m.K) (▲) and 0.59 W/(m.K) (●).

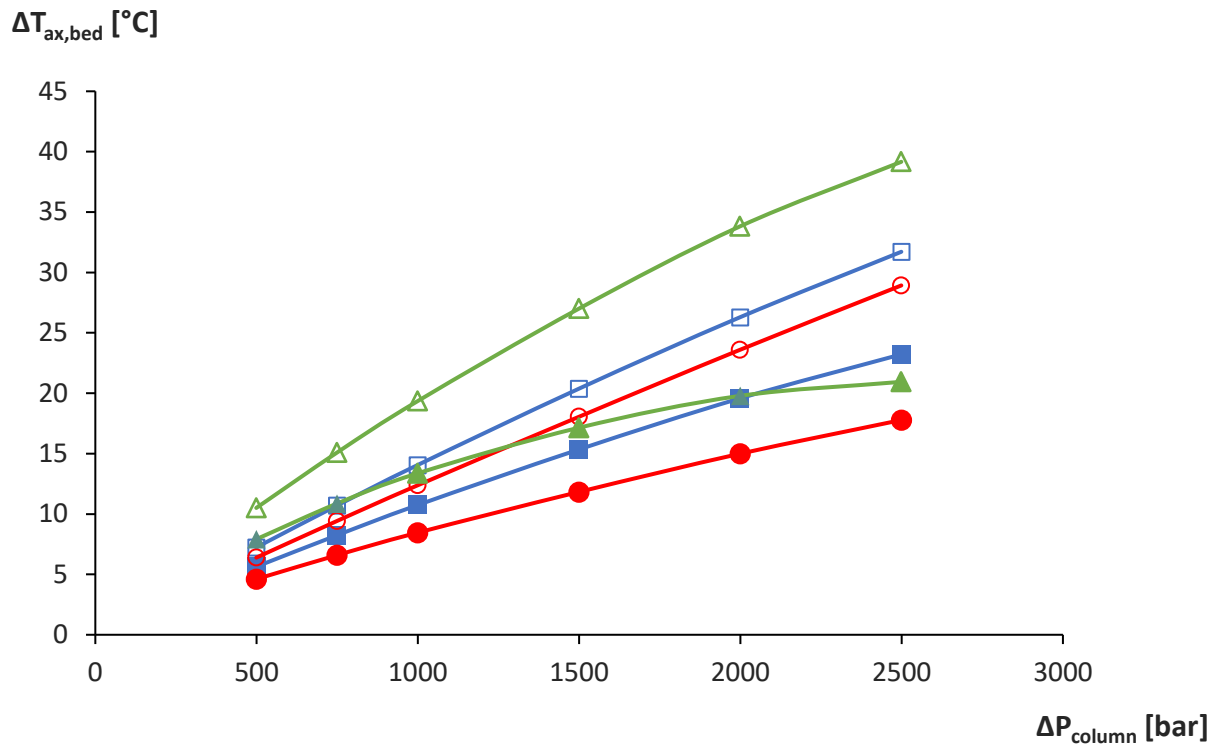


Figure S4. Evolution of $\Delta T_{ax,bed}$ as a function of the column pressure drop for natural convection (full symbols) and adiabatic (open symbols) boundary conditions. Considered liquids: pure water (■), pure ACN (▲) and a 40/60(v%/v%) ACN/H₂O mixture (●). (same data as in figure 9, different representation)