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1	Coupling of chiral and achiral stationary phases in supercritical fluid chromatography: evaluating
2	and improving retention prediction
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22 Abstract

23 Isomers and stereoisomers are always challenging to separate. Column coupling may provide 24 improved chromatographic selectivity, necessary for the separation of the compounds with similar 25 chemical and structural properties. The relatively low viscosity of supercritical fluids, used as mobile 26 phases allows for the coupling of several columns in series in supercritical fluid chromatography 27 (SFC), without exceeding the pressure limits of the system. The aim of this study is to propose 28 reliable prediction of the retention behaviour of analytes on a coupled column system, based on a 29 limited number of initial analyses. The chiral compounds atenolol, ephedrine, propranolol, 30 mianserin, labetalol and nadolol, besides the diastereomers guinine and guinidine, and the 31 structural isomers of aminophenol and aminocresol were used as model analytes. The retention 32 behaviour of the analytes was determined on the individual chiral columns Lux Cellulose-1, Lux 33 Cellulose-2, Lux Cellulose-3, Lux Cellulose-4, Lux Amylose-2 and the achiral columns Luna NH₂, Luna 34 Silica, Synergi RP and FluoroSep RP. The mobile phase was composed of CO_2 mixed with 20% (v/v) 35 MeOH, which contained 0.1% (v/v) trifluoroacetic acid and 0.1% (v/v) isopropylamine. The retention 36 factors of the analytes on coupled stationary phases were predicted, and subsequently compared to 37 the experimentally obtained ones. Relative deviations of predicted and experimental retention 38 factors were in range from 0.00% to 51.91%. Flow rate and back pressure of the screening 39 conditions were adjusted to improve prediction precision on four column combinations, with varying 40 success rates. The average relative deviations of retention factors were reduced to 2.84% - 6.59% by 41 adjusting flow rate, and to 2.30% - 8.57% by adjusting back pressure. The most successful approach, 42 flow rate adjustment, was then applied to select a column combination providing improved 43 resolution of the structurally similar components of silymarin extract. 44 Keywords: supercritical fluid chromatography, column coupling, chiral and achiral separations,

45 retention prediction

47 **1. Introduction**

Supercritical fluid chromatography (SFC) is a separation technique developed in the 1960s 48 49 with an initial commercial growth in the 1980s [1,2]. The major benefits of the technique are 50 attributed to the properties of supercritical fluids. SFC primarily uses CO_2 as mobile phase, which has 51 a critical pressure and temperature of 73.8 bar and 31.1°C, respectively [3]. Thanks to the lower 52 viscosity and better diffusion properties in comparison to LC, higher velocities in the column can be 53 used [4]. This leads to an increased throughput as well as a reduced analysis time and organic 54 solvent consumption, compared to conventional HPLC. These features of SFC are beneficial also in 55 terms of organic solvent cost and waste reduction, being an important asset in the pharmaceutical 56 industry, while they present a reduced impact on the environment as well [5].

57 The benefits for enantioseparation were recognised early on in SFC development, with the 58 first separation of enantiomers by SFC reported in 1985 [6]. Nowadays, SFC is well established for 59 the separation of chiral compounds [7]. Chiral method development is a labour intensive and time-60 consuming process, mainly based on a trial-and-error approach, given the unpredictability and 61 complexity of enantioselectivity [8]. Because of the properties of the stationary phase, which are the 62 main determinants of retention behaviour and enantioselectivity of the analytes, its choice is the 63 most crucial step in method development [9].

64 The low viscosity of the mobile phase, responsible for the low pressure drop across the 65 column, and the compatibility of the same mobile phase with a variety of stationary phases are the 66 main features allowing column coupling in SFC. Several methods have been published taking 67 advantage of the properties of SFC for column coupling [10–22]. Berger and Wilson [10] coupled 68 several columns with the same stationary phase, achieving high efficiencies on a 2.2 m long column 69 for the separation of polycyclic aromatic hydrocarbons. Phinney et al. [11] demonstrated the use of 70 coupled chiral and achiral stationary phases for the separation of structurally related β-blockers and 71 1,4-benzodiazepines. In the work of Lesellier et al. [15,16] coupling of two to seven stationary phases

72 was explored to separate the isomers of β -carotene. Serial coupling of silica and diol columns 73 improved the selectivity for glycolipids in the work of Deschamps et al. [17]. Barnhart et al. [18] used 74 the chiral columns Chiralpak AD-H and Chiralcel OD-H, coupled in tandem, to separate a mixture of 75 four in-house synthesized stereoisomers. Adequate separation of the four stereoisomers was not 76 achieved using a single chiral column, but the individual column results indicated the possibility of 77 the separation on a coupled column system [18]. Welch et al. [20] developed a column screening 78 tool for tandem column coupling to separate complex or multicomponent chiral mixtures. In other 79 studies, the advantages of column coupling were used for impurity profiling [12], for steroid profiling 80 [13], for the stereoisomeric separation of potential anti-diabetic compounds [14], to separate 81 spirostanol diastereomers [22], and when the separation on a single column was found insufficient 82 [19,21].

83 When coupling columns in series, the column positioned as first is operating under a higher pressure. This causes an increase in density of the mobile phase, and consequently of its solvation 84 85 power, resulting in a decrease in retention compared to the column positioned downstream. The 86 importance of column back pressure as optimization parameter, and of column sequence in a 87 coupled column system was highlighted in the work of Wang et al. [23,24]. This, as well as the 88 advantage of coupling columns with different selectivities, was demonstrated on the separation of 89 stereoisomers of molecules with more than one chiral centre [23,24]. The importance of pressure in 90 a coupled-columns system was further illustrated in respective works of Lynen's group [25,26]. For 91 the prediction of an optimal column combination to separate an achiral and chiral compounds 92 mixture, they used the stationary-phase optimized selectivity supercritical fluid chromatography 93 methodology (SOS-SFC)[25,26], where a selection of stationary phases is made using retention 94 factors from a screening step. The columns are then connected with zero-dead-volume couplers. 95 Correspondence between predictions and experiments improved when the same average pressure 96 was used in all experiments by adapting the back pressure or the flow rate.

97 In the actual study, five chiral and four achiral conventional columns are screened under 98 generic chromatographic conditions for their ability to separate enantiomers and/or structural 99 isomers. Combinations of coupled columns, a total of 60 systems coupled by PEAK couplers, were 100 screened under the same conditions. The aim of this work is to define a stationary-phase selection 101 approach in the method-optimization process in real lab conditions, maximizing the chance that the 102 most effective column combinations are chosen for the separation of given mixtures. The stationary-103 phase selection is based on a prediction approach with the highest correspondence to the 104 experimental results. The effects of flow rate and back pressure changes are explored to further 105 improve the prediction ability of the employed model. The findings are then applied to select the 106 best column combination for the separation of silymarin, an extract from the Silybum marianum 107 plant. This mixture consists of six structurally similar flavonolignans, silybin A (SB A), silybin B (SB B), isosilybin A (ISB A), isosilybin B (SB B), silychristin (SCH), silydianin (SD), and the flavonoid taxifolin 108 109 (TX).

110 2. Material and methods

111 2.1 Chemicals and reagents

112 The following compounds were used as a model sample set (Figure S-1): (R)-(+)-atenolol (99%), (S)-113 (-)-atenolol (99%), (1R,2S)-(-)-ephedrine (99%), quinine (99%), labetalol, nadolol, o-aminophenol 114 (99%), m-aminophenol, p-aminophenol, 2-amino-m-cresol (96%), 2-amino-p-cresol (97%), and 5-115 amino-o-cresol (97%), all purchased from Sigma-Aldrich (St. Louis, MO, USA), (1S,2R)-(+)-ephedrine 116 (99%), (R)-propranolol (99%), (S)-propranolol (99%), and quinidine (90%) obtained from Fluka 117 Chemie (Neu-Ulm, Switzerland), (S)-(+)-mianserin and (R)-(-)-mianserin were gifts. CO₂ quality 118 4.5 (purity ≥ 99.995%) was from Messer (Sint-Pieters-Leeuw, Belgium). HPLC grade methanol 119 was provided by VWR Chemicals (Fontenay-sous-Bois, France). Trifluoroacetic acid (99%) and 120 isopropylamine (\geq 99.5%) were purchased from Aldrich (Steinheim, Germany).

The standards of SB B (≥95%), ISB A (≥95%), SCH (≥95%), SD (≥95%), and TX (≥85%) were
purchased from Sigma-Aldrich (Steinheim, Germany). SB (≥92.53%) was obtained from HWI
Analytik (Rülzheim, Germany). ISB B and SD were provided by PhytoLab (Vestenbergsgreuth,
Germany).

125 2.2 Stationary phases

- 126 Chiral stationary phases Lux Cellulose-1 (LC-1), Lux Cellulose-2 (LC-2), Lux Cellulose-3 (LC-3),
- 127 Lux Cellulose-4 (LC-4), and Lux Amylose-2 (LA-2) were from Phenomenex (Utrecht, The
- 128 Netherlands). The dimensions of all chiral columns were 250 mm x 4.6 mm i.d. with 3 μm
- 129 particle size. Achiral columns Luna NH2 (LNH2), Luna Silica (LS), both with dimensions 100 mm
- 130 x 4.6 mm and particle size 3μ m, and Synergi Polar RP (SRP), 100 mm x 4.6 mm with 4 μ m
- particle size, were also obtained from Phenomenex. The achiral column FluoroSep-RP (FSRP)
- 132 Phenyl with dimensions 150 mm x 4.6 mm and 3 μm particle size was acquired from ES
- 133 Industries (West Berlin, Germany).

134 **2.3 Instruments**

- 135 All analyses were performed on an SFC Acquity Ultra Performance Convergence Chromatography
- 136 (UPC²) system from Waters (Milford, MA, USA). The system is equipped with a binary pump, an
- autosampler with a fixed loop of 10 µL thermostated at 10°C, a convergence manager, a photo diode
- 138 array (PDA) detector, a back pressure regulator and an external column oven. For data collection and
- 139 processing, the Empower software (3 V7.10, 2010, Waters) was used.

140 **2.4 Preparation of standard and test solutions**

- 141 Stock solutions for each individual compound were prepared by dissolving the appropriate amount
- 142 of test set standard in 10 mL MeOH to achieve a concentration of 0.5 mg/mL (or 1 mg/mL for
- aminocresol). Standard solutions of each individual compound were prepared by diluting the stock
- solutions with MeOH to achieve final concentrations of 0.25 mg/mL (0.20 mg/mL for aminocresol).

Mixture solutions of structural of positional isomers of the same compound were prepared by
transferring appropriate amounts of stock solutions into a 10 mL test tube, leading to final
concentrations of 0.25 mg/mL for each analyte (concentrations of aminophenol and aminocresol
were 0.17 and 0.20 mg/mL, respectively).

Stock solutions of flavonolignans of the silymarin complex and of TX were prepared by dissolving the
appropriate amount of the standard in MeOH to achieve a concentration of 1 mg/mL (or 2 mg/mL
for SB). Test solutions were prepared by diluting 100 μL stock solution with MeOH to achieve 100
μg/mL (200 μg/mL for SB). A mixture solution was prepared by adding 100 μL of each standard
solution to 400 μL MeOH.

154 **2.5 Chromatographic screening conditions**

For the screening experiments, generic chromatographic conditions from an earlier defined chiral
separation strategy were used [27,28]. All screening experiments were performed in isocratic mode
at a constant flow rate of 3.0 mL/min, with a mobile phase consisting of 20% MeOH, containing 0.1%
(v/v) isopropylamine and 0.1% (v/v) trifluoroacetic acid, combined with 80% CO₂. The back pressure
was kept at 150 bar and the column temperature at 30°C, while the temperature of the sample
compartment was 10°C. The injection volume was 5 µL and the detection wavelength was 220 nm.

161 **2.6 Calculations**

The test set analytes were initially screened on the individual chiral and achiral columns using the above screening conditions. To predict the retention factors of analytes in coupled column systems, three equations were employed and compared. Eq. 1 was designed to calculate the retention factor of a given compound on serially connected stationary phases in liquid chromatography [29]. Eq. 2 was an update for SFC to account for the void time [30], while Eq. 3 is based on a similar expression used in gas chromatography [31].

168

169
$$k_{A,B} = \frac{\phi_A k_A + \phi_B k_B}{\phi_A + \phi_B}$$
 (Eq. 1) [29]

170
$$k_{A,B} = \frac{\phi_A k_A t_{0A} + \phi_B k_B t_{0B}}{\phi_A t_{0A} + \phi_B t_{0B}}$$
 (Eq. 2) [30]

171
$$k_{A,B} = k_A t_{0A} + k_B t_{0B}$$
 (Eq. 3) [31]

where $k_{A,B}$ is the predicted retention factor of an analyte in a coupled column system, k_A and k_B are the experimental retention factors on the individual stationary phases, Φ_A and Φ_B represent the

174 column lengths of the stationary phases (see section 2.2), while t_{OA} and t_{OB} are the void times of the

175 individual stationary phases.

176 The experimental retention factors of the analytes on a given column are calculated using the

equation $k = (t_R - t_0)/t_0$, where t_R stands for the retention time of the analyte and t_0 the void time,

178 which was determined as the retention time of the first system peak on the chromatogram. The

equation $t_R = kt_0 + t_0$ was used to calculate retention times predicted on coupled columns from

180 predicted retention factors and predicted void times.

181 Correspondence between calculated (k_{cal}) and experimentally obtained retention factors (k_{exp}) values 182 was assessed by the relative deviation (RD) [32]:

183
$$RD(\%) = 100 \frac{|k_{cal} - k_{exp}|}{(k_{cal} + k_{exp})/2}$$
 (Eq. 4)

184 3. Results and discussion

185 **3.1. Column screening and selection of prediction equations**

All model compounds were analysed on the chiral and achiral columns under the initial screening conditions. The same chromatographic conditions (Section 2.5) were applied for screening of the coupled column systems. All combinations coupling chiral - chiral (CH-CH), chiral - achiral (CH-ACH) and achiral - chiral (ACH-CH) columns were tested. The combinations of achiral and achiral stationary phases were not examined.

191 The isomers were analysed first separately to determine the elution order of the isomers, and then 192 as a mixture. Values of k, t_R , and t_0 obtained from analysis of the mixtures were used in the 193 calculations.

To assess the best way to predict the retention factors of analytes on coupled columns, three equations (Eqs. 1-3) were applied. The estimated retention factors were compared to the experimental on the coupled columns. The best prediction model should provide the smallest difference between predicted and experimental results. Averages and ranges of relative deviations for categories of coupled combinations are listed in Table 1.

The least analogy between predicted and experimental retention factors of analytes on coupled columns was achieved using Eq. 3. The equation predicts retention factors from the individual retention factors and void times from single column screening, without considering the lengths of the columns. The average relative deviations for ACH-CH and CH-ACH systems, which were 35 – 40 cm long, were 27.1% and 33.0%, respectively. The average relative deviation for 50 cm long CH-CH systems was 59.2%. From the relative deviations obtained using Eq. 3 (Table 1A), it can be concluded that the deviation increased with the length of the coupled columns.

206 Better but still poor prediction ability was obtained when using Eq. 2. This equation considers

207 retention factors, column lengths and void times. Opposed to the Eq.3, the best agreement between

208 predicted and experimental values was on the CH-CH coupled systems (average RD = 6.40%), which

are the longest coupled systems. However, this equation performed insufficiently well for the

210 systems where shorter achiral columns were connected to chiral columns, with average relative

211 deviations of 12.0% and 17.8% for the ACH-CH and CH-ACH systems, respectively.

Eq. 1 provided the best prediction ability, with average relative deviations of 7.3%, 9.8%, and 6.7%

for ACH-CH, CH-ACH, and CH-CH coupled systems, respectively, with 20 examined systems in each

- 214 category. This equation does not include the void time values of the individual columns for
- 215 prediction of retention factors on the coupled systems.

216 The estimated void time of serially coupled columns was expressed as the sum of the single-column 217 void times. Using predicted retention factors and void times, the retention times of the analytes 218 were predicted and compared to the experimental values. The average relative deviations of 219 predicted and experimental retention times were 5.0%, 5.9% and 5.1%, respectively, for ACH-CH, 220 CH-ACH and CH-CH coupled systems (Table 1B). On average, the agreement between predicted and 221 experimental retention times is better than that of retention factors, where average relative 222 deviations are higher, providing promising results for retention time predictions in coupled systems. 223 Depending on its position in the coupled system, the column operates under different pressures. 224 Using the same column pair in different sequences may result in differences in the prediction quality 225 (Fig. 1). An example of such difference is seen for the combination of LA-2 and FSRP columns. When 226 the achiral column FSRP is first in the column sequence, the average relative deviation was 8.4%. 227 When FSRP is placed after the LA-2 column, the % RD increased to 39.2% (Fig. 1 A, B). The retention 228 of analytes in the first column decreases as a consequence of the higher pressure and mobile phase 229 density. Experimental retention factors were closer to the predicted when the FSRP column was 230 placed in first position, where the column had a less impact on the separation, compared to the LA-2 231 column connected as second in the sequence. 232 Another example is the combination of the LS and LC-3 columns (Fig. 1 A, B). Placing the chiral 233 column at the beginning of the system resulted in an average deviation of 3.2%, while having the 234 shorter achiral column first, leads to an increase in the deviation to 10.4%. An example of a 235 difference in a combination of two chiral columns with the same length is LC-1 and LC-3 (Fig. 1 C). 236 Upstream connection of LC-1 led to an %RD of 16.4%. The deviation decreased to 6.65% when LC-1

is coupled downstream in the system. In general, changing the sequence in CH-CH systems caused

smaller differences in relative deviations as opposed to when achiral and chiral columns were

239 coupled in series. This may be linked to the lower efficiencies of the chiral columns compared to the

240 achiral ones and different column lengths of achiral columns.

241 In four occasions, the effect of the pressure difference under which columns operate, resulted in 242 such significant shift between predicted and experimental results, that the elution sequence was not 243 predicted correctly anymore. This was the case for the closely eluting isomers of aminocresol on the 244 LC-3+LC-4 system and for aminophenol on the LC-2+LC-1, LC-1+SRP, and LS+LC-1 systems. In all 245 these examples, at least two of the three isomers co-eluted on one of the columns during single 246 column screening. Because the retention behaviour of co-eluting compounds was unclear in the 247 single column, it was more difficult to predict the retention of the compounds in the coupled 248 systems, even if relative deviations between predicted and experimental retention factors were 249 relatively small.

In the LS column, *o*- and *m*- aminophenol eluted as one peak, followed by *p*-aminophenol. The
elution sequence of the aminophenol isomers on the LC-1 column was *o*-, *p*-, and *m*-aminophenol.
Based on the results from the single column screening, the predicted elution sequence was the same
as the observed on the LC-1 chiral column. The experimental elution sequence, however, was as
follows: *o*-aminophenol, *m*-aminophenol, *p*-aminophenol, with deviations 1.1%, 33.6%, and 4.8%,
respectively. The difference in retention times between the first and last eluting analytes was again
less than one minute.

257 Overall, it was possible to predict the correct elution sequence of analytes in a majority of these 258 preliminary experiments (60 systems x 7 analytes). As the density of the mobile phase is of major 259 importance in SFC, the pressure differences complicate the retention behaviour prediction of the 260 analytes. Inconsistency in the mobile phase density as well as the pressure drops in single columns 261 and coupled column systems contributed to large deviations between predicted and measured 262 retention factors. This is especially important when analysing samples where analytes are eluting 263 closely to each other. Therefore, some adjustments of the chromatographic conditions were tested 264 in order to improve the prediction performance.

265 **3.2 Adjustment of chromatographic conditions**

266 So far, all the individual and coupled columns were screened under the same chromatographic 267 conditions. Back pressure and flow rate were kept constant, which led to differences in the 268 generated system pressure and therefore in mobile-phase properties. To minimize these differences, 269 flow rate and back pressure were adjusted to generate the same average pressure in the individual 270 column and the coupled system. The average pressure was calculated as the average of 271 backpressure and system pressure. The effectiveness of this approach was demonstrated on four 272 column combinations: LS+LA-2, LC-3+SRP, LA-2+FSRP, LC-2+LC-3, representing ACH-CH, CH-ACH and 273 CH-CH couplings and covering the range of screened columns. The selected column combinations 274 varied both in length of the system and in particle sizes, as well as in their averages and ranges of % RD. 275

276 3.2.1 Flow rate adjustment

277 To improve the agreement between experimental and predicted values, the flow rate was adapted 278 to generate the same system pressure on the coupled columns as obtained during the screening of 279 single columns. The flow rate was also adjusted during the individual column screening step, when 280 columns of different lengths were coupled, so that the generated system pressure, and 281 consequently the average pressure, was the same throughout all experiments. For example, the 282 system pressures generated with 3 mL/min flow rate were 239, 256, and 311 bar for LS, LA-2, and 283 LS+LA-2, respectively. In the actual approach, the flow rate of LA-2 was adjusted to 2.66 mL/min to 284 generate 239 bar system pressure resulting in an average pressure of 194.5 bar. The retention 285 factors on the coupled system were predicted using the results from LS (3mL/min) and LA-2 (2.66 286 mL/min) individual column screenings. The predicted retention factors were then compared to the 287 experimental on LS+LA-2, analysed at a flow rate of 1.89 mL/min, which also generated 239 bar 288 system pressure. This approach of flow rate adjustment was applied on the four above-mentioned 289 coupled-column systems using equation 1 to calculate predicted retention factors (see Table 2).

For this approach, the void time was calculated using Eq. 5 to predict the retention time in a coupledsystem:

292
$$t_{03} = \frac{t_{01} F R_1 + t_{02} F R_2}{F R_3}$$
 (Eq. 5)

where t_{01} and t_{02} are the void times of the individual stationary phases, FR_1 and FR_2 the flow rates on the individual stationary phases, FR_3 the flow rate on the coupled-column system, and t_{03} the void time of the coupled-column system.

296 The first examined combination was the ACH-CH system with LS and LA-2. The pressure generated 297 on LA-2 using the general screening conditions at 3 mL/min was 256 bar (average pressure 203 bar). 298 To reach the same system pressure, the LS column had to be operated at an increased flow rate of 299 3.40 mL/min. A flow rate of 2.20 mL/min generated the same pressure on the coupled column 300 system. The predicted retention factors were compared to the experimental as % RD obtained at 301 adjusted conditions (Table 2). The average deviation for retention factors (Table 2) and retention 302 times (Table 3) decreased from 10.61% and 8.2% for initial results, to 6.3% and 3.4%, respectively. 303 The same trend was seen on this system when flow rates were adjusted to generate the system 304 pressure of 239 bar (average pressure 194.5 bar) generated as described at the beginning of this 305 section. The average and maximal deviations of retention factors (Table 2) in the LS+LA-2 system 306 decreased from 10.6% and 21.1% to 6.6% and 16.4%, respectively. The retention times (Table 3) also 307 resembled the predicted values better when applying a flow rate change, as opposed to the initial 308 predictions, decreasing from 8.2% to 4.9% average RD. 309 The experimental retention factor of o-aminophenol correlated the worst for the system at 2.20

mL/min flow rate (average pressure 203 bar), with a deviation of 22.6%. Here the maximum RD was
higher than for the initial prediction, which was 21.1% and originated from 2-amino-*p*-cresol (Table
2). However, the maximal % RD for the retention times decreased from 17.1% to 11.5% and 10.7%,
depending on the applied flow rate (Table 3).

314 The flow rate was changed similarly for the three other column combinations to generate the same 315 system pressure both on the individual columns and on the coupled system. For the CH-ACH 316 combination with LC-3 and SRP, the flow rate was changed to 1.79 mL/min for the chiral column and 317 1.55 mL/min for coupled system to match the system pressure on the achiral column. On the other 318 hand, to reach the system pressure of the LC-3 column analysed at 3 mL/min, the SRP column 319 requires a flow rate higher than 4 mL/ml, which was above the operating limits of the equipment. 320 For this reason, the LC-3 and SRP combination was analysed with just one alternative flow rate, 1.55 321 mL/min. The highest deviations of retention factors in the original predictions were seen for p-322 aminophenol and 5-amino-o-cresol with 41.8% and 44.3, respectively. By reducing the flow rate on 323 the LC-3+SRP system, the correspondence between predicted and experimental retention factors of 324 the analytes improved significantly to 9.8% for p-aminophenol and 0.8% for 5-amino-o-cresol. This 325 contributed substantially to the improvement in average and maximal deviations, which decreased 326 from 9.1% and 44.3% in the initial prediction, to 2.8% and 9.8%, respectively (Table 2). Retention 327 time predictions also improved considerably from average RD of 5.6% to 1.1% (Table 3). 328 The system pressures on LC-2 and LC-3 at general screening conditions were the same. Therefore, 329 the coupled system was also analysed only once, with the flow rate reduced to 1.93 mL/min, as 330 opposed to the initial flow rate of 3.0 mL/min. The relative deviations for the system LC-2+LC-3 were

not as high as in the previous example. The largest deviation in retention factor was 22.6%, caused

by *p*-aminophenol. By changing the flow rate of the coupled system, the retention of the analyte

factors, where the maximal deviation improved to 18.4%. As in the previous system, the retention
behaviour of the analytes was better predicted, with average deviation of retention factors 5.3% for

increased. This resulted in a better correspondence between predicted and experimental retention

and LC-2+LC-3, compared to the original average RD of 8.88% (Table 2). Also for retention times a

reduction in average RD was seen from 6.8% to 3.5% (Table 3).

14

338 Using the screening conditions with changed flow rate, the most substantial improvement in average 339 and maximal % RD values of retention factors and retention times was seen for the LA-2 and FSRP 340 coupled system. During the screening step, the LA-2 and FSRP columns generated system pressures 341 of 256 and 266 bar, respectively. To predict the retention factors, LA-2 was analysed with a flow rate 342 of 3.16 mL/min generating a system pressure of 266 bar. A system pressure of 256 bar was achieved 343 at 2.80 mL/min flow rate on the FSRP column. The LA-2+FSRP system was analysed with reduced 344 flow rates 1.88 and 2.04 mL/min. The most significant improvement seen was in maximal % RD for 345 the retention factor, which was reduced from 51.9% to 21.0% and 19.7%, respectively. The maximal 346 51.9% RD was in the initial prediction attributed to the ephedrine enantiomers. By using the 347 adjusted flow rate strategy, correlation with the predicted retention times improved drastically, with 348 RD 3.0% and 6.7% for these compounds at flow rates 1.88 and 2.04 mL/min, respectively. The 349 maximal deviation of retention factors under adjusted flow rates was caused by aminocresol 350 isomers. In general, the average retention factors deviations improved considerably from 39.2% to 351 6.6% and 4.4% for LA-2+FSRP system analysed with flow rates 1.88 and 2.04 mL/min, respectively. 352 For the retention times, average deviations decreased from 9.4% to 3.8% and 2.4%. 353 The strategy of changing the flow rates according to a generated system pressure provided better 354 results for all four column combinations relative to the initial predictions. The elution sequences of 355 the analytes were predicted correctly for each mixture, and the average % RD of the predicted 356 retention factors and retention times improved. By keeping the same back pressure of 150 bar for all 357 experiments and changing the flow rate depending on the generated system pressure, a smaller 358 pressure difference was created between the column inlet and outlet. The smaller pressure drop 359 allowed a more consistent mobile phase density than when a flow rate of 3 mL/min was used, 360 regardless of the length of a column and the stationary phase character.

The retention factors were also predicted using Equations 2 and 3, to determine whether also better results from the flow rate change were reached. Using Eq. 3, there was still a considerable deviation

363 between predicted and experimental retention factors. The lowest average RD was 16.1% for the 364 LS+LA-2 system analysed at 2.20 mL/min. The average deviation between experimental and 365 predicted retention factors, calculated using Eq. 2, i.e. 4.8% for LA-2+FSRP at 2.04 mL/min flow rate 366 and 5.4% for LC-2+LC-3 at 1.93 mL/min flow rate (Table S-1), is comparable with the results obtained 367 using Eq. 1. However, when the predictions were made for shorter coupled systems, the deviations 368 became larger. The highest average deviation was 23.3%, obtained for the LS+LA-2 system, when analysed at 2.20 mL/min flow rate (Table S-1). Because of this inconsistency in results related to the 369 370 different column lengths and the high deviations for shorter coupled systems, Eq. 1 provides the 371 most reliable predictions.

372 3.2.2 Back pressure adjustment

A second approach tested towards reducing the % RD for retention factors and retention times of analytes was adjusting the back pressure in the coupled-column systems. The back pressure was manipulated, to ensure that the average pressure inside the columns was the same during the screening of the individual columns, and in the coupled systems to achieve consistency in the mobile phase density during the experiments. The % RD values of retention factors and retention times predicted with adapted back pressure were then compared with the predictions from the original experiments (Tables 2 and 3).

For the ACH-CH combination, the LS+LA-2 system was again tested. During individual column screening with flow rate 3 mL/min and back pressure 150 bar, the system pressure generated on the LS column was 239 bar. The average of these pressures in the column was therefore 194.5 bar. To simulate the same average pressure, the LA-2 column was screened at 141 bar back pressure, which generated a system pressure of 248 bar (corresponding to an average pressure of 194.5 bar). To achieve the same average column pressure as on the individual columns, the LS+LA-2 system was tested at 118 bar back pressure, which generated a system pressure of 271 bar.

From the individual LS column and LA-2 column screening experiments with reduced back pressure, the retention factors on the LS+LA-2 system were predicted using Eq. 1. The predicted values were compared to the experimental retention factors obtained by analysing the coupled system with 118 bar back pressure. Improved correlation of predicted and experimental retention factors was observed, reducing average and maximal deviations from 10.6% and 21.1%, to 6.33% and 18.1%, respectively (Table 2).

393 The system pressure reached by screening the LA-2 column under back pressure 150 bar was 256 394 bar, making the average column pressure 203 bar. To accommodate the average pressure generated 395 by the LA-2 column during initial screening, LS was screened at 159 bar and the LS+LA-2 system at 396 125 bar. Overall, average retention factor deviation improved from 10.6% for the initial predictions, 397 to 6.8%. However, in this system with slightly increased average pressure, the maximal deviation 398 was larger (24.0%) than for the initial predictions (21.1%) (Table 2). The deviation of 24.0% was 399 caused by a prolonged retention of o-aminophenol, increasing the difference between predicted and 400 experimental retention factor values.

The three other coupled-systems, LC-3+SRP, LA-2+FSRP, and LC-2+LC-3, were tested by the same 401 402 approach. To simulate the average column pressure of 179 bar, generated in the SRP column with 403 150 bar back pressure, the LC-3 column and LC-3+SRP system were analysed with back pressures of 404 123 and 115 bar, respectively. In general, agreement between predicted and experimental values 405 improved, with average RD for retention factors 3.6% and for retention times 0.9%. However, only a 406 slight improvement was detected for the LC-2+LC-3 system. Both chiral columns generated the same 407 system pressure under initial screening conditions, i.e. 262 bar. To match the average pressure of 408 206 bar, the LC-2+LC-3 system was analysed with a back pressure of 113 bar, generating a system 409 pressure of 299 bar. In this system, the maximum relative deviation for the retention factors 410 increased from 22.6% to 29.7%. The worst agreement between predicted and experimental 411 retention factor values was found for labetalol. Moreover, the elution sequence of the aminocresol

412 isomers was predicted incorrectly. The analytes eluted in the sequence: 5-amino-o-cresol, 2-amino-

413 *m*-cresol, and 2-amino-*p*-cresol, but the predicted order was 2-amino-*m*-cresol, 5-amino-*o*-cresol,

414 and 2-amino-*p*-cresol, with relative deviations 3.0%, 17.6%, and 11.8%, respectively.

415 The LA-2+FSRP system was analysed similarly, changing the back pressures to 145 and 154 bar to 416 screen the FSRP and LA-2 columns, respectively. To equal the average pressures on the columns 417 during the screening process, the LA-2+FSRP system was analysed with 112 bar and 117 bar back 418 pressures. In both cases, the back pressure adjustment led to significant improvements in the 419 agreement between predicted and experimental retention factors and retention times. The average 420 deviation of retention factors decreased from 39.2% for the initial predictions, to 5.3% and 2.3% 421 when analysed with 112 bar and 117 bar back pressures, respectively. 422 In general, this approach provided improved similarity between predicted and experimental results 423 of the analytes, comparable to the results obtained by the flow rate adapting strategy. An exception 424 is the LC-2+LC-3 system, where the average % RD of the retention factors and retention times

425 improved only slightly, and the maximum % RDs were higher than the initial predictions. Moreover,

426 the elution sequence of the aminocresol isomers, eluting within less than a minute of each other,

427 was incorrectly predicted on the LC-2+LC-3 system. This inconsistency in the results when applying

428 the back pressure change strategy may be caused by the various pressure drops generated in

429 columns of different lengths.

While system and back pressures on the columns were the same for the adjusted flow rate approach (for instance LS, LA-2, LS+LA-2 all had a back pressure of 150 bar and a generated system pressure of 256 bar), this was not the case for the back pressure adjusting approach, where the aim was to reach the same average pressure of 203 bar on the columns and system. For example, the LA-2 column was analysed with a back pressure of 150 bar, generating a 256 bar system pressure, while the LS column was analysed at 159 bar and a system pressure of 247 bar. The LS+LA-2 coupled system had

436 a back pressure of 125 bar, generating a system pressure of 281 bar, leading to altered pressure437 drops on the columns.

Because in all abovementioned cases the average column pressure was the same (203 bar), different
pressures were affecting the mobile phase properties in the columns. The differences between inlet
and outlet pressures of the columns may be the cause of inconsistencies in mobile phase densities,
and consequently the reason for the inferior accuracy (average RD = 6.8%) of retention behaviour
predictions than in the flow rate change approach (average RD = 6.3%).

The highest deviations were mostly observed for aminophenol or aminocresol analytes. These are the only compounds from the sample set with one benzene ring and amino and hydroxyl groups. The density of the mobile phase might affect the ionization of these analytes in a non-linear way, making their behaviour even more difficult to predict.

447 **3.3 Application to silymarin**

448 The strategy of changing the operating flow rate to achieve consistency in average column pressure 449 was applied on the screening and selection of chiral and achiral columns for the separation of the 450 silymarin complex compounds. Silymarin compounds were strongly retained on the chiral stationary 451 phases, exceeding analysis time over 60 min when using the mobile phase with 20% (v/v) of 452 modifier. Therefore, the modifier content was increased to 40% (v/v) to keep the analysis time 453 below 30 min. The aim was to screen all columns in this study for silymarin separation and the 454 screening started with the SRP column. Based on the earlier experiments, this column was known to 455 generate the lowest system pressure due to the 4 μ m particles and the 10 cm length. When the 456 screening was started with another column, the generated system pressure may be too high and 457 impossible to reach on the SRP column, because of the flow rate limit of the instrument. The system pressure generated on the column at flow rate 3.5 mL/min was 237 bar. The flow rate of the method 458 459 was adjusted for each column depending on the system pressure generated on the SRP column (for 460 example the flow rate on LNH2 was 2.56 mL/min, and on LC-3 was 1.82 mL/min).

None of the screened individual columns provided separation of all seven tested silymarin
compounds (data not shown). Analysis on achiral columns LS, LNH2, SPRP, and FSRP provided two,
four, two, and one peak, respectively. The chiral columns provided a more efficient separation, with
six peaks on the chromatograms of LC-1, LC-2, LC-3, and LC-4. SCH and SD co-eluted on LC-1, SB A
and ISB A on LC-2, SB A and ISB B on LC-3, and ISB A and ISB B on LC-4. Compounds SB A and SB B,
and ISB A and ISB B co-eluted on the LA-2 column, resulting in five observed peaks.

467 From the screening results, the retention factors and subsequently the selectivities on each column 468 combination were estimated using Eq. 1. Based on the predicted retention factors, the combination 469 of columns LC-2 and LC-3 was chosen as the most promising for the separation of the silymarin 470 compounds. Screening on the individual LC-2 column showed co-elution of SB A and ISB A, while on 471 the individual column LC-3 column compounds SB A and ISB B remained unseparated (Fig. 2A, B). 472 Two coupled-column systems, LC-2+LC-3 and LC-3+LC-2 (Fig. 2 C, D), were tested for separation of 473 silymarin. The predicted retention factors and retention times were compared to the experimentally 474 obtained ones (Table S-2). The relative deviations of retention factors ranged from 49.44% for SB B

to 67.14% for TX, both in LC-3+LC-2 system. Similarly, the predicted void times, 2.96 and 2.91 min for

476 LC-2+LC-3 and LC-3+LC-2 systems, respectively, were very different from the experimental values:

477 4.63 min for LC-2+LC-3 and 4.59 min for LC-3+LC-2 systems. This drastic increase in deviations,

478 compared to the previous experiments, can be explained by different compressibility of the mobile

479 phase with 40% modifier, resulting in different volumes of the mobile phase in the system. However,

480 the agreement between predicted retention times, calculated from predicted retention factors and

481 predicted void times, and experimental retention times was very good, with % RD not exceeding 7%

482 (Table S-2).

483 Even if the agreement between predicted and experimental retention factors was poor, the

484 combination of the two columns, LC-2 and LC-3, did show at least partial separation of all tested

485 compounds (Fig. 2 C, D). Baseline separation was not achieved on either system, but the best

resolution for the critical peak pair, i.e. ISB B/SB A, was achieved when the LC-3 column coupled first
(Fig. 2D). Further optimisation of the separation was not examined since this was not the purpose of
the actual study.

489 4. Conclusions

490 Compounds with similar chemical structures and properties are always challenging to separate, 491 requiring a unique chromatographic selectivity, which may be created by column coupling. Chiral 492 and achiral columns were tested individually and serially coupled in SFC, to provide an insight into 493 the retention behaviour of the test analytes in such coupled-column systems. Three equations were 494 employed to predict the retention factors of analytes on the coupled systems, based on individual 495 column retentions. The highest similarity between the predicted and experimental retention factors 496 was seen for Eq. 1., with an average RD for all coupled systems (ACH-CH, CH-ACH, and CH-CH) of 497 7.96%, compared to 12.07% and 39.77% for Eq. 2 and Eq. 3, respectively (Table 1).

498 Instrumental conditions, i.e. flow rate and back pressure, were adjusted to achieve the same 499 average column pressure for single-column and coupled-system analysis, and consequently to 500 improve predictability. The results of the study show that while a change in back pressure may 501 provide faster analysis, the drawback of this approach is the higher inconsistency between the 502 experimental and predicted results when compared to the flow rate change approach. With back 503 pressure adjustment, an increased maximal % RD in coupled systems and an unreliable predictions 504 of the elution sequence was demonstrated on closely eluting compounds. An explanation is that the 505 tested Equation 1 was designed to be used primarily in LC where the properties of a mobile phase do 506 not change throughout the column. Hence, they do not account for changes in mobile-phase 507 properties caused by pressure variation.

The strategy of flow-rate change, generating a constant system pressure during screening as well as after coupling, showed to be more reliable. The problems related to pressure drops were minimized by keeping the back pressure constant, while generating the same system pressure. The

511 improvement in % RD was demonstrated on the four examined coupled systems: LS+LA-2, LC-3+SRP,
512 LA-2+FSRP and LC-2+LC-3.

The applicability of this approach was demonstrated on the separation of seven structurally related compounds of the silymarin complex. Based on the results from the screening, the most suitable columns were selected. While baseline separation of all compounds was not achieved on single columns, an improved resolution of the critical peak pairs was observed when a coupled column system was used.

518 In conclusion, it is recommended to start the single column screening with the lowest system

519 pressure generating column, to ensure that the generated system pressure is achievable on the

520 other tested columns. The used flow rate should be selected with respect to the operating limits of

the instrument and the column. The generated system pressure should then be matched on each

522 single column by adjusting the flow rate. Finally, the flow rate should be adjusted on the coupled

523 systems to generate the same system pressure as achieved on the single columns.

524 Conflict of Interest

525 The authors declare no conflict of interest.

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528 References

- 529 [1] D.A. Klesper, E., Corwin, A.H. Turner, High Pressure Gas Chromatography above Critical
 530 Temperatures, J. Org. Chem. 27 (1962) 700–701.
- 531 [2] T.L. Chester, The Role of Supercritical Fluid Chromatography in Analytical Chemistry, J.
- 532 Chromatogr. Sci. 24 (1986) 226–229.
- 535 [4] W. Majewski, E. Valery, O. Ludemann-Hombourger, Principle and applications of supercritical
 536 fluid chromatography, J. Liq. Chromatogr. Relat. Technol. 28 (2005) 1233–1252.
- 537 [5] C.J. Welch, W.R. Leonard, J.O. DaSilva, M. Biba, J. Albaneze-Walker, D.W. Henderson, B. Laing,
- 538 D.J. Mathre, Preparative chiral SFC as a green technology for rapid access to enantiopurity in 539 pharmaceutical process research, LC-GC Eur. 18 (2005) 264–272.
- 540 [6] P.A. Mourier, E. Eliot, M.H. Caude, R.H. Rosset, A.G. Tambute, Supercritical and Subcritical
- 541 Fluid Chromatography on a Chiral Stationary Phase for the Resolution of Phosphine Oxide 542 Enantiomers, Anal. Chem. 57 (1985) 2819–2823.
- 543 [7] Y. Zhao, G. Woo, S. Thomas, D. Semin, P. Sandra, Rapid method development for chiral
 544 separation in drug discovery using sample pooling and supercritical fluid chromatography-
- 545 mass spectrometry, J. Chromatogr. A. 1003 (2003) 157–166.
- 546 [8] M. K. Mone, K.B. Chandrasekhar, Evaluation of generic gradients, sample pooling and MS
- 547 detection as chiral resolution screening strategies on diverse chiral stationary phases,
- 548 Chromatographia. 73 (2011) 985–992.
- 549 [9] C. Galea, D. Mangelings, Y. Vander Heyden, Characterization and classification of stationary
 550 phases in HPLC and SFC a review, Anal. Chim. Acta. 886 (2015) 1–15.

- 551 [10] T.A. Berger, W.H. Wilson, Packed Column Supercritical Fluid Chromatography with 220 000
 552 Plates, Anal. Chem. 65 (1993) 1451–1455.
- K.W. Phinney, L.C. Sander, S.A. Wise, Coupled Achiral/Chiral Column Techniques in Subcritical
 Fluid Chromatography for the Separation of Chiral and Nonchiral Compounds, Anal. Chem. 70
 (1998) 2331–2335.
- 556 [12] C. West, E. Lemasson, S. Bertin, P. Hennig, E. Lesellier, Interest of achiral-achiral tandem
 557 columns for impurity profiling of synthetic drugs with supercritical fluid chromatography, J.
 558 Chromatogr. A. 1534 (2018) 161–169.
- 559 [13] J. Teubel, B. Wüst, C.G. Schipke, O. Peters, M.K. Parr, Methods in endogenous steroid
- 560 profiling A comparison of gas chromatography mass spectrometry (GC–MS) with
- 561 supercritical fluid chromatography tandem mass spectrometry (SFC-MS/MS), J. Chromatogr.

562 A. 1554 (2018) 101–116.

- 563 [14] A. Akchich, J. Charton, E. Lipka, Application of tandem coupling of columns in supercritical
- 564 fluid chromatography for stereoisomeric separation: Optimization and simulation, J.
- 565 Chromatogr. A. 1588 (2019) 115–126.
- 566 [15] E. Lesellier, K. Gurdale, A. Tchapla, Separation of cis/trans isomers of β-carotene by
 567 supercritical fluid chromatography, J. Chromatogr. A. 844 (1999) 307–320.
- E. Lesellier, C. West, A. Tchapla, Advantages of the use of monolithic stationary phases for
 modelling the retention in sub/supercritical chromatography: Application to cis/trans-β carotene separation, J. Chromatogr. A. 1018 (2003) 225–232.
- 571 [17] F.S. Deschamps, E. Lesellier, J. Bleton, A. Baillet, A. Tchapla, P. Chaminade, Glycolipid class
 572 profiling by packed-column subcritical fluid chromatography, J. Chromatogr. A. 1040 (2004)
 573 115–121.
- 574 [18] W.W. Barnhart, K.H. Gahm, S. Thomas, S. Notari, D. Semin, J. Cheetham, Supercritical fluid

575 chromatography tandem-column method development in pharmaceutical sciences for a 576 mixture of four stereoisomers, J. Sep. Sci. 28 (2005) 619-626. 577 A.J. Alexander, A. Staab, Use of achiral/chiral SFC/MS for the profiling of isomeric [19] 578 cinnamonitrile/hydrocinnamonitrile products in chiral drug synthesis, Anal. Chem. 78 (2006) 579 3835-3838. 580 [20] P.H. C. Welch, M. Biba, J.R.Gouker, G. Kath, P. Augustine, Solving multicomponent chiral 581 separation challenges using a new SFC tandem column screening tool, Chirality. 19 (2007) 184-189. 582 583 [21] C. Brunelli, Y. Zhao, M.H. Brown, P. Sandra, Development of a supercritical fluid 584 chromatography high-resolution separation method suitable for pharmaceuticals using 585 cyanopropyl silica, J. Chromatogr. A. 1185 (2008) 263–272. 586 [22] Y. Zhao, J. McCauley, X. Pang, L. Kang, H. Yu, J. Zhang, C. Xiong, R. Chen, B. Ma, Analytical and 587 semipreparative separation of 25 (R/S)-spirostanol saponin diastereomers using supercritical 588 fluid chromatography, J. Sep. Sci. 36 (2013) 3270–3276. 589 C. Wang, Y. Zhang, Effects of column back pressure on supercritical fluid chromatography [23] 590 separations of enantiomers using binary mobile phases on 10 chiral stationary phases, J. 591 Chromatogr. A. 1281 (2013) 127–134. 592 [24] C. Wang, A.A. Tymiak, Y. Zhang, Optimization and simulation of tandem column supercritical 593 fluid chromatography separations using column back pressure as a unique parameter, Anal. 594 Chem. 86 (2014) 4033-4040. 595 [25] S. Delahaye, F. Lynen, Implementing stationary-phase optimized selectivity in supercritical 596 fluid chromatography, Anal. Chem. 86 (2014) 12220–12228.

597 [26] R.S. Hegade, F. Lynen, Chiral stationary phase optimized selectivity supercritical fluid

598 chromatography: A strategy for the separation of mixtures of chiral isomers, J. Chromatogr.

599 A. 1586 (2019) 116–127.

- 600 [27] K. De Klerck, G. Parewyck, D. Mangelings, Y. Vander Heyden, Enantioselectivity of
- 601 polysaccharide-based chiral stationary phases in supercritical fluid chromatography using
- 602 methanol-containing carbon dioxide mobile phases, J. Chromatogr. A. 1269 (2012) 336–345.
- 603 [28] K. De Klerck, Y. Vander Heyden, D. Mangelings, Generic chiral method development in
- supercritical fluid chromatography and ultra-performance supercritical fluid chromatography,
 J. Chromatogr. A. 1363 (2014) 311–322.
- 606 [29] S. Nyiredy, Z. Szűcs, L. Szepesy, Stationary phase optimized selectivity liquid chromatography:
- 607 Basic possibilities of serially connected columns using the "PRISMA" principle, J. Chromatogr.
- 608 A. 1157 (2007) 122–130.
- [30] I. Salsinha, C. Galea, D. Mangelings, Y. Vander Heyden, Coupling columns in Supercritical Fluid
 Chromatography and retention prediction under isocratic conditions The 'PRISMA' model
 application, Unpublished results.
- 612 [31] C.F. Poole, S.K. Poole, Chromatography today, fifth ed., Elsevier Science Publishers,
- 613 Amsterdam, The Netherlands, 1991.
- 614 [32] Y. Grooten, P. Riasová, I. Salsinha, D. Mangelings, Y. Vander Heyden, Stationary-phase
- optimized selectivity in supercritical fluid chromatography using a customized Phase
- 616 OPtimized Liquid Chromatography kit: comparison of different prediction approaches, Anal.
- 617 Bioanal. Chem. 412 (2020) 6553–6565.

619 Figure captions

- 620 **Figure 1:** Relative deviations between experimental and predicted (using Eq.1) retention factors for
- 621 specific combinations of (A) ACH-CH, (B) CH-ACH and (C) CH-CH coupled systems, depicting average
- 622 (bars), minimum and maximum (dots) RD (%). Column abbreviations: see text.
- 623 Figure 2: Chromatograms of silymarin separation on (A) LC-2 and (B) LC-3 individual columns, and on
- 624 (C) LC-2+LC-3, and (D) LC-3+LC-2 coupled column systems, using the general screening conditions
- 625 with 40% MeOH and adjusted flow rate to generate a system pressure of 237 bar. TX taxifolin, SCH
- 626 silychristin, ISB B isosilybin B, SB A silybin A, ISB A isosilybin A, SD silydianin, SB B silybin B.

Table 1: (A) Relative deviations (%) between predicted and experimental retention factors using the
three different equations, and (B) relative deviations (%) of retention times (in min) estimated from
retention factors and void time (see section 2.6).

631 A

			Α	CH-CH			CH-A	СН		CH-CH				
		Avg. RD (%)	Med. RD (%)	Min. RD (%)	Max. RD (%)	Avg. RD (%)	Med. RD (%)	Min. RD (%)	Max. RD (%)	Avg. RD (%)	Med. RD (%)	Min. RD (%)	Max. RD (%)	
	Eq. 1	7.29	6.47	0.07	50.69	9.83	6.38	0.04	51.91	6.76	4.74	0.00	33.59	
	Eq. 2	12.03	8.98	0.01	50.19	17.79	14.55	0.01	56.34	6.40	4.51	0.01	29.57	
	Eq. 3	27.09	26.46	1.18	53.55	33.05	28.84	1.21	81.40	59.19	58.19	31.19	82.09	
632		1				I								

633 B	
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		(%) (%) (%) (%					CH-A	СН		CH-CH				
_		RD	RD	RD	RD	RD	RD	RD	Max. RD (%)	RD	RD	Min. RD (%)	RD	
	Eq. 1	4.99	3.91	0.02	24.91	5.88	4.51	0.07	22.24	5.05	3.32	0.03	27.92	
621		1				I								

634

636 Table 2: Relative deviations (|% RD|) obtained when applying Eq. 1, for retention factors at the initial conditions (korg), after the flow rate adjustment (kFR), and

637 after the back pressure (\mathbf{k}_{BP}) adjustment. AP – average pressure.

	LS+LA-2							LC-3+SRP			LC-2+LC-3			LA-2+FSRP				
AP (bar)		2	.03		194	4.5		17	79		20	06		20	03		208	
RD (%)	k org	k fr	k _{вр}	k org	k fr	k _{вр}	k org	k fr	k _{вр}	k org	k fr	k _{вр}	k org	k fr	k _{вр}	k org	k fr	kвр
(R)-(+)-atenolol	17.57	1.13	1.57	17.57	7.47	7.08	5.75	1.18	3.87	13.62	4.58	4.21	37.64	7.69	5.66	37.64	0.24	1.81
(S)-(-)-atenolol	17.57	1.13	1.57	17.57	7.47	7.08	5.75	1.18	0.56	13.50	4.46	4.73	37.64	7.69	5.66	37.64	0.24	1.81
(1R,2S)-(-)-ephedrine	12.10	2.65	2.70	12.10	6.07	5.92	6.65	1.74	0.57	6.31	1.23	5.22	51.91	3.02	0.45	51.91	6.72	0.18
(1S,2R)-(+)-ephedrine	12.10	2.65	2.70	12.10	6.07	5.92	6.65	1.74	0.57	5.40	3.22	5.27	51.91	3.02	0.45	51.91	6.72	0.18
(R)-propranolol	9.75	0.47	1.81	9.75	1.85	1.34	0.44	4.73	3.15	6.59	0.50	2.82	39.30	5.10	1.98	39.30	4.80	0.21
(S)-propranolol	9.75	0.47	1.81	9.75	1.85	1.34	3.66	4.73	1.00	6.67	1.01	3.05	39.30	5.10	1.98	39.30	4.80	0.21
(S)-(+)-mianserine	10.57	0.21	1.90	10.57	1.67	1.31	6.18	2.55	3.60	12.42	2.26	1.85	40.21	5.18	4.20	40.21	0.47	2.91
(R)-(-) mianserine	10.57	0.21	1.90	10.57	1.67	1.31	6.70	2.93	4.20	12.63	3.17	3.07	40.21	5.18	4.20	40.21	0.47	2.91
quinine	9.43	0.48	1.73	9.43	0.14	0.64	7.39	0.20	4.80	16.21	6.02	5.97	48.73	9.59	8.67	48.73	1.30	6.54
quinidine	9.49	12.41	0.48	9.49	0.81	0.82	4.88	3.37	7.98	13.78	3.58	3.76	45.56	10.50	9.34	45.56	0.98	5.91
labetalol (1)	1.09	10.36	12.77	1.09	11.62	10.28	0.20	4.60	5.95	3.58	16.85	29.74	38.20	1.93	2.37	38.20	3.90	0.23
labetalol (2)	1.09	10.36	12.77	1.09	11.62	10.28	0.20	4.60	5.95	4.80	8.49	21.49	38.20	1.93	2.37	38.20	3.90	0.23
labetalol (3)	1.09	10.36	12.77	1.09	11.62	10.28	0.20	4.60	5.95	4.80	8.49	21.49	38.20	1.93	2.37	38.20	3.90	0.23
labetalol (4)	1.09	10.36	12.77	1.09	11.62	10.28	0.20	4.60	5.95	4.80	8.49	21.49	38.20	1.93	2.37	38.20	3.90	0.23
nadolol (1)	11.61	5.41	6.29	11.61	2.29	2.87	3.66	2.26	3.82	9.14	0.97	0.50	39.06	3.14	0.97	39.06	1.26	0.08
nadolol (2)	11.61	5.41	6.29	11.61	2.29	2.87	3.66	2.26	3.82	7.54	1.40	0.35	39.06	3.14	0.97	39.06	1.26	0.08
nadolol (3)	14.20	3.22	4.05	14.20	4.89	5.45	3.66	2.13	3.82	13.07	1.68	0.09	40.03	3.14	0.97	40.03	2.34	0.08
nadolol (4)	14.20	3.22	4.05	14.20	4.89	5.45	3.66	2.13	3.82	11.37	2.13	0.47	40.03	3.14	0.97	40.03	2.34	0.08
o-aminophenol	13.42	22.58	23.95	13.42	16.41	18.14	10.49	0.58	2.47	6.27	9.84	19.79	26.17	9.54	8.45	26.17	11.33	7.65
m-aminophenol	5.24	3.17	4.58	5.24	1.25	0.01	16.35	0.58	2.47	2.52	4.01	5.05	40.65	1.70	1.94	40.65	2.82	2.07
p-aminophenol	19.84	6.44	4.97	19.84	16.20	14.48	41.76	9.75	2.03	22.58	18.36	12.87	49.48	1.70	2.59	49.48	13.10	3.48
2-amino-m-cresol	9.95	10.10	9.94	9.95	13.35	12.90	9.57	2.44	4.16	1.25	7.29	17.64	35.13	21.00	19.70	35.13	19.73	15.34
2-amino-p-cresol	21.14	10.10	9.94	21.14	12.02	12.58	26.21	2.51	4.02	4.44	5.43	11.83	23.29	21.00	19.70	23.29	4.24	1.36
5-amino-o-cresol	10.20	18.49	19.81	10.20	2.58	3.18	44.25	0.78	2.80	9.85	4.38	3.01	23.29	21.00	19.70	23.29	4.24	1.36
Average RD (%)	10.61	6.31	6.80	10.61	6.57	6.33	9.09	2.84	3.64	8.88	5.33	8.57	39.22	6.59	5.33	39.22	4.37	2.30
Min. RD (%)	1.09	0.21	0.48	1.09	0.14	0.01	0.20	0.20	0.56	1.25	0.50	0.09	23.29	1.70	0.45	23.29	0.24	0.08
Max. RD (%)	21.14	22.58	23.95	21.14	16.41	18.14	44.25	9.75	7.98	22.58	18.36	29.74	51.91	21.00	19.70	51.91	19.73	15.34

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	I		10.					C 2 . CDI					I					
				LA-2			L	.C-3+SRF		l	LC-2+LC-3			LA-2+FSRP				
AP (bar)		20)3		19	4.5		17	79		20	06		20)3		20	08
RD (%)	t R ORG	t _{R FR}	t _{r bp}	t r org	t _{R FR}	t _{r bp}	tr org	t r fr	t _{r bp}	t r org	t _{R FR}	t _{r bp}	t r org	t _{R FR}	t _{r bp}	t r org	t _{R FR}	t _{r bp}
(R)-(+)-atenolol	17.06	0.87	1.65	17.06	8.06	9.01	3.74	1.22	0.33	13.80	5.08	5.66	13.14	6.04	5.69	13.14	0.89	5.78
(S)-(-)-atenolol	17.06	0.87	1.65	17.06	8.06	9.01	3.74	1.22	1.78	14.14	5.11	6.28	13.14	6.04	5.69	13.14	0.89	5.78
(1R,2S)-(-)-ephedrine	8.76	0.39	1.52	8.76	5.04	6.24	2.94	0.30	1.89	3.62	0.45	0.10	7.86	2.11	2.07	7.86	3.45	4.66
(1S,2R)-(+)-ephedrine	8.76	0.39	1.52	8.76	5.04	0.80	2.94	0.30	1.89	3.66	0.50	0.47	7.86	2.11	2.07	7.86	3.45	4.66
(R)-propranolol	8.05	2.11	1.88	8.05	2.80	3.77	1.66	1.39	0.63	5.46	0.57	0.04	9.64	3.74	2.99	9.64	3.40	4.70
(S)-propranolol	8.05	2.11	1.88	8.05	2.80	3.77	3.02	1.39	2.48	5.61	0.24	0.15	9.64	3.74	2.99	9.64	3.40	4.70
(S)-(+)-mianserine	9.76	1.67	1.60	9.76	2.89	3.90	4.61	0.64	0.10	10.71	2.61	3.21	11.45	4.00	4.40	11.45	0.46	6.31
(R)-(-) mianserine	9.76	1.67	1.60	9.76	2.89	3.90	4.57	0.70	0.02	10.37	3.17	4.02	11.45	4.00	4.40	11.45	0.46	6.31
quinine	8.14	2.13	1.87	8.14	1.57	2.52	4.38	0.63	0.05	12.87	5.21	6.09	12.43	5.93	6.43	12.43	1.39	7.96
quinidine	8.43	9.98	2.66	8.43	1.12	2.39	3.51	0.74	1.46	10.82	3.37	4.42	12.86	6.81	7.15	12.86	0.20	7.86
labetalol (1)	1.34	5.46	6.05	1.34	6.56	4.33	1.38	1.61	1.05	0.97	10.49	18.75	10.02	0.25	0.47	10.02	1.56	4.72
labetalol (2)	1.34	5.46	6.05	1.34	6.56	4.33	1.38	1.61	1.05	4.61	4.92	13.21	10.02	0.25	0.47	10.02	1.56	4.72
labetalol (3)	1.34	5.46	6.05	1.34	6.56	4.33	1.38	1.61	1.05	4.61	4.92	13.21	10.02	0.25	0.47	10.02	1.56	4.72
labetalol (4)	1.34	5.46	6.05	1.34	6.56	4.33	1.38	1.61	1.05	4.61	4.92	13.21	10.02	0.25	0.47	10.02	1.56	4.72
nadolol (1)	11.10	2.35	1.88	11.10	3.45	5.19	2.75	0.16	0.50	8.08	1.61	1.44	12.39	2.89	2.50	12.39	1.53	4.64
nadolol (2)	11.10	2.35	1.88	11.10	3.45	5.19	2.75	0.16	0.50	7.04	1.94	1.54	12.39	2.89	2.50	12.39	1.53	4.64
nadolol (3)	13.13	0.67	0.15	13.13	5.50	7.22	2.75	1.55	0.50	11.25	2.17	1.74	13.04	2.89	2.50	13.04	2.21	4.64
nadolol (4)	13.13	0.67	0.15	13.13	5.50	7.22	2.75	1.55	0.50	10.21	2.56	2.18	13.04	2.89	2.50	13.04	2.21	4.64
o-aminophenol	5.35	10.74	10.40	5.35	7.61	7.35	7.54	0.35	0.49	2.52	5.30	10.89	0.13	3.58	2.11	0.13	4.54	0.99
m-aminophenol	5.13	0.01	0.33	5.13	2.39	2.94	11.05	0.35	0.49	0.34	1.89	1.69	6.91	1.72	2.83	6.91	2.09	5.58
p-aminophenol	13.93	5.62	5.92	13.93	11.46	11.72	16.73	5.40	1.11	11.05	9.12	5.11	11.58	1.72	3.14	11.58	7.24	6.27
2-amino-m-cresol	3.00	3.53	2.30	3.00	5.40	3.88	6.95	0.76	0.51	0.62	3.68	9.47	2.48	8.84	7.25	2.48	8.37	2.56
2-amino-p-cresol	9.31	3.53	2.30	9.31	5.15	4.21	17.04	0.89	0.56	1.52	2.68	6.06	1.55	8.84	7.25	1.55	1.30	3.93
5-amino-o-cresol	3.53	8.47	8.08	3.53	0.17	1.10	22.24	0.28	0.52	4.87	1.87	0.08	1.55	8.84	7.25	1.55	1.30	3.93

2.36

0.20

8.37

4.98

0.99

7.96

Table 3: Relative deviations (% RD) obtained when applying Eq. 1, for retention times (min) at the initial conditions (t_{R ORG}), after the flow rate adjustment (t_R 639 $_{FR}$), and after the back pressure adjustment (t_{RBP}). AP – average pressure. 640

Max. RD (%) 641

Average RD (%)

Min. RD (%)

8.25

1.34

17.06

3.42

0.01

10.74

3.14

0.15

10.40

8.25

1.34

17.06

4.86

0.17

11.46

4.94

0.80

11.72

5.55

1.38

22.24

1.10

0.16

5.40

0.85

0.02

2.48

6.81

0.34

14.14

3.52

0.24

10.49

5.38

0.04

18.75

9.36

0.13

13.14

3.78

0.25

8.84

3.57

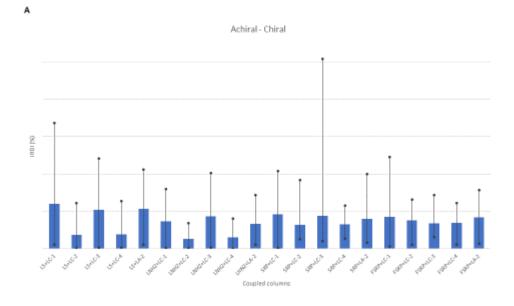
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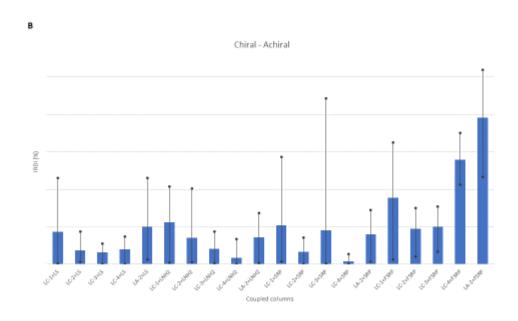
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9.36

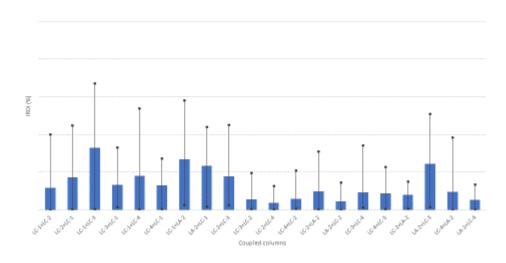
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13.14





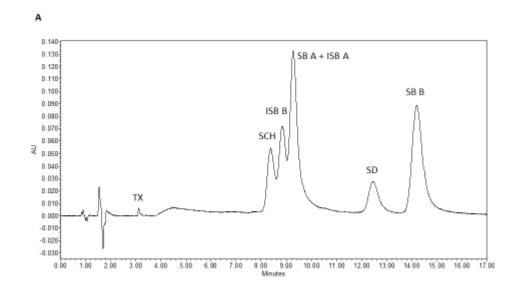
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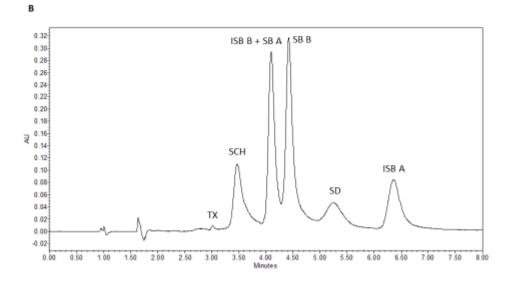


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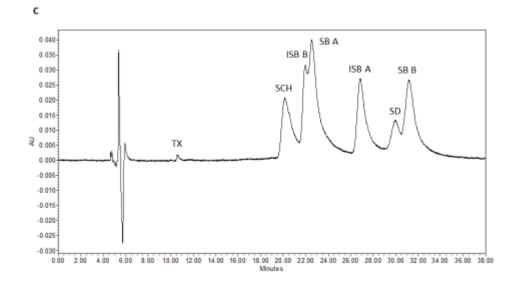
646 Figure 2

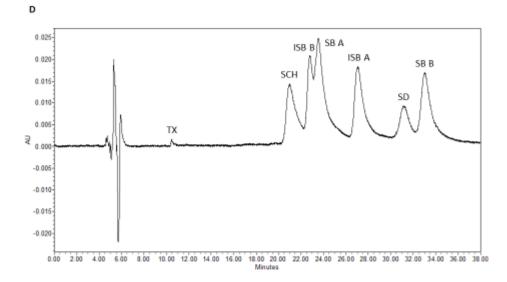
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652	Supplementary material
653	Coupling of chiral and achiral stationary phases in supercritical fluid chromatography: evaluating
654	and improving retention prediction
655	
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673 Figure captions

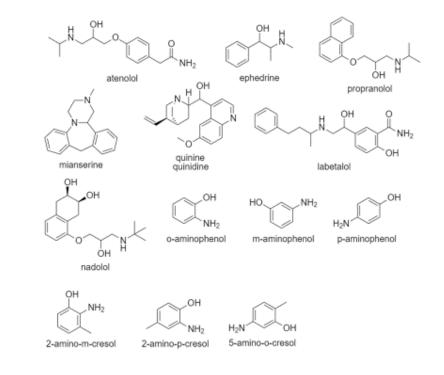


Figure S-1: Chemical structures of the sample set compounds.

- 676 **Table S-1:** Relative deviations (% RD) obtained applying Eq. 2, for (A) retention factors and (B)
- 677 retention times (min), after the flow rate adjustment, and at the initial conditions.

	LS+	LA-2	LC-3+SRP	LC-2+LC-3	LA-2-	FSRP		
FR (mL/min)	2.20	1.89	1.55	1.93	1.88	2.04		
SP (bar)	256	239	208	262	256	266		
(A) Retention	factors after	flow rate adj	justment					
Average RD (%)	23.27	21.01	16.57	5.38	9.24	4.80		
Min. RD (%)	5.36	2.65	4.66	0.24	2.20	0.43		
Max. RD (%)	43.20	38.32	58.81	19.44	20.00	13.29		
Retention factors	at initial cond	litions						
Average RD (%)	12	.73	16.47	7.16	35.21			
Min. RD (%)	0.	89	2.07	1.05	21.81			
Max. RD (%)	27	.68	52.40	23.66	53.30			
(B) Retention	time after fl	ow rate adjus	stment					
Average RD (%)	13.57	12.39	6.74	2.98	5.57	2.97		
Min. RD (%)	1.26	0.12	1.25	0.08	1.76	0.50		
Max. RD (%)	26.95	24.92	21.25	11.85	12.41	6.90		
Retention times a	t initial condi	tions						
Average RD (%)	6.	49	6.70	5.57	11	.85		
Min. RD (%)	0.	63	0.10	0.66	0.	64		
Max. RD (%)	12	.64	26.76	11.62	85	85.38		

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679 FR – flow rate, SP – system pressure

Table S-2: Evaluation and overview of predicted and experimental retention factors (k) and retention
 times (t_R) for silymarin compounds for (A) LC-2+LC-3 and (B) LC-3+LC-2 coupled systems.

(A)	Predicted		Experimental			
	k	t _R (min)	k	RD (%)	t _R (min)	RD (%)
Silybin A	7.01	23.72	3.86	57.88	22.54	5.12
Silybin B	10.26	33.36	5.72	56.79	31.16	6.80
Isosilybin A	8.23	27.35	4.80	52.79	26.86	1.80
Isosilybin B	6.74	22.93	3.75	57.15	21.99	4.17
Silychristin	6.11	21.04	3.35	58.31	20.16	4.31
Silydianin	9.62	31.46	5.48	54.89	30.02	4.67
Taxifolin	2.58	10.59	1.29	66.75	10.60	0.08
(B)	Predicted		Experimental			
	k	t _R (min)	k	RD (%)	t _R (min)	RD (%)
Silybin A	7.01	23.28	4.12	51.90	23.52	1.01
Silybin B	10.26	32.74	6.19	49.44	33.04	0.92
Isosilybin A	8.23	26.84	4.89	50.92	27.06	0.81
Isosilybin B	6.74	22.51	3.97	51.84	22.81	1.34
Silychristin	6.11	20.65	3.58	52.28	21.01	1.71
Silydianin	9.62	30.87	5.80	49.59	31.22	1.12